

ORGANIC COMPOSITIONS

Benefit of Pending Applications

This application claims the benefit of pending commonly assigned
5 provisional patent applications 60/294864 filed May 30, 2001; 60/350187 filed
January 15, 2002; 60/350557 filed January 22, 2002; 60/353011 filed
January 30, 2002; 60/376219 filed April 29, 2002, and 60/378424 filed May
7, 2002, incorporated herein in their entirety.

Field of The Invention

The present invention relates to semiconductor devices, and in particular,
to semiconductor devices having an organic low dielectric constant material and
processes for the manufacture thereof.

Background of The Invention

In an effort to increase the performance and speed of semiconductor
devices, semiconductor device manufacturers have sought to reduce the
linewidth and spacing of interconnects while minimizing the transmission losses
and reducing the capacitive coupling of the interconnects. One way to
20 diminish power consumption and reduce capacitance is by decreasing the
dielectric constant (also referred to as "k") of the insulating material, or
dielectric, that separates the interconnects. Insulator materials having low
dielectric constants are especially desirable, because they typically allow faster
signal propagation, reduce capacitance and cross talk between conductor lines,
25 and lower voltages required to drive integrated circuits.

Since air has a dielectric constant of 1.0, a major goal is to reduce the
dielectric constant of insulator materials down to a theoretical limit of 1.0, and

several methods are known in the art for reducing the dielectric constant of insulating materials. These techniques include adding elements such as fluorine to the composition to reduce the dielectric constant of the bulk material. Other methods to reduce k include use of alternative dielectric material matrices. 5 Another approach is to introduce pores into the matrix.

Therefore, as interconnect linewidths decrease, concomitant decreases in the dielectric constant of the insulating material are required to achieve the improved performance and speed desired of future semiconductor devices. For example, devices having interconnect linewidths of 0.13 or 0.10 micron and 10 below seek an insulating material having a dielectric constant (k) < 3 .

Currently silicon dioxide (SiO_2) and modified versions of SiO_2 , such as fluorinated silicon dioxide or fluorinated silicon glass (hereinafter FSG) are used. These oxides, which have a dielectric constant ranging from about 3.5-4.0, are commonly used as the dielectric in semiconductor devices. While SiO_2 and FSG 15 have the mechanical and thermal stability needed to withstand the thermal cycling and processing steps of semiconductor device manufacturing, materials having a lower dielectric constant are desired in the industry.

Methods used to deposit dielectric materials may be divided into two categories: spin-on deposition (hereinafter SOD) and chemical vapor deposition 20 (hereinafter CVD). Several efforts to develop lower dielectric constant materials include altering the chemical composition (organic, inorganic, blend of organic/inorganic) or changing the dielectric matrix (porous, non-porous). Table I summarizes the development of several materials having dielectric constants ranging from 2.0 to 3.5. (PE = plasma enhanced; HDP = high-density plasma) 25 However, the dielectric materials and matrices disclosed in the publications shown in Table 1 fail to exhibit many of the combined physical and chemical properties desirable and even necessary for effective dielectric materials, such as higher mechanical stability, high thermal stability, high glass transition temperature, high modulus or hardness, while at the same time still being able 30 to be solvated, spun, or deposited on to a substrate, wafer, or other surface.

Therefore, it may be useful to investigate other compounds and materials that may be used as dielectric materials and layers, even though these compounds or materials may not be currently contemplated as dielectric materials in their present form.

Table 1

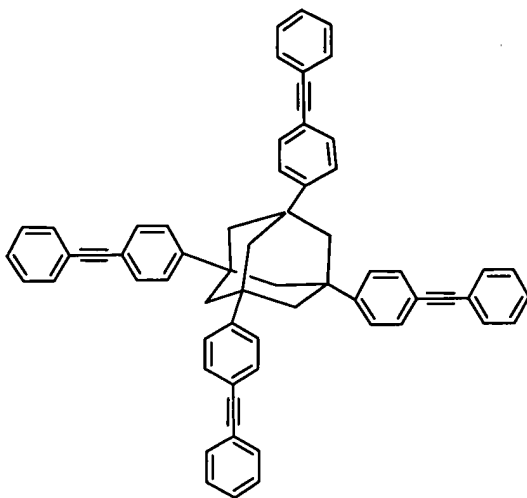
MATERIAL	DEPOSITION METHOD	DIELECTRIC CONSTANT (k)	REFERENCE
Fluorinated silicon oxide (SiOF)	PE-CVD; HDP-CVD	3.3-3.5	US Patent 6,278,174
Hydrogen Silsesquioxane (HSQ)	SOD	2.0-2.5	US Patents 4,756,977; 5,370,903; and 5,486,564; International Patent Publication WO 00/40637; E.S. Moyer et al., "Ultra Low k Silsesquioxane Based Resins", <i>Concepts and Needs for Low Dielectric Constant <0.15 μm Interconnect Materials: Now and the Next Millennium</i> , Sponsored by the American Chemical Society, pages 128-146 (November 14-17, 1999)
Methyl Silsesquioxane (MSQ)	SOD	2.4-2.7	US Patent 6,143,855
Polyorganosilicon	SOD	2.5-2.6	US Patent 6,225,238
Fluorinated Amorphous Carbon (a-C:F)	HDP-CVD	2.3	US Patent 5,900,290
Benzocyclobutene (BCB)	SOD	2.4-2.7	US Patent 5,225,586
Polyarylene Ether (PAE)	SOD	2.4	US Patents 5,986,045; 5,874,516; and 5,658,994
Parylene (N and F)	CVD	2.4	US Patent 5,268,202
Polyphenylenes	SOD	2.6	US Patents 5,965,679 and 6,288,188B1; and Waeterloos et al., "Integration Feasibility of Porous SiLK Semiconductor Dielectric", Proc. Of the 2001 International Interconnect Tech. Conf., pp. 253-254 (2001).
Thermosettable benzocyclobutenes, polyarylenes, thermosettable perfluoroethylene monomer	SOD	2.3	International Patent Publication WO 00/31183
Poly(phenylquinoxaline), organic polysilica	SOD	2.3-3.0	US Patents 5,776,990; 5,895,263; 6,107,357; and 6,342,454; and US Patent Publication 2001/0040294
Organic polysilica	SOD	Not reported	US Patent 6,271,273
Organic and inorganic Materials	SOD	2.0-2.5	US Patent 6,156,812
Organic and inorganic Materials	SOD	2.0-2.3	US Patent 6,171,687
Organic materials	SOD	Not reported	US Patent 6,172,128
Organic	SOD	2.12	US Patent 6,214,746
Organosilsesquioxane	CVD, SOD	< 3.9	WO 01/29052
Fluorosilsesquioxane	CVD, SOD	< 3.9	WO 01/29141

Unfortunately, numerous organic SOD systems under development with a dielectric constant between 2.0 and 3.5 suffer from certain drawbacks in terms of mechanical and thermal properties as described above; therefore a need exists in the industry to develop improved processing and performance for dielectric films in this dielectric constant range. In addition, industry demands

materials having demonstrated low dielectric constant extendibility, i.e. capable of being reduced to an even lower dielectric constant, e.g., from 2.7 to 2.5 to 2.2 to 2.0 and below.

5 Reichert and Mathias describe compounds and monomers that comprise adamantane molecules, which are in the class of cage-based molecules and are taught to be useful as diamond substitutes. (Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1993, Vol. 34 (1), pp. 495-6; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1992, Vol. 33 (2), pp. 144-5; Chem. Mater., 1993,
10 Vol. 5 (1), pp. 4-5; Macromolecules, 1994, Vol. 27 (24), pp. 7030-7034; Macromolecules, 1994, Vol. 27 (24), pp. 7015-7023; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1995, Vol. 36 (1), pp. 741-742; 205th ACS National Meeting, Conference Program, 1993, pp. 312; Macromolecules, 1994, Vol. 27 (24), pp. 7024-9; Macromolecules, 1992, Vol. 25 (9), pp. 2294-306;
15 Macromolecules, 1991, Vol. 24 (18), pp. 5232-3; Veronica R. Reichert, PhD Dissertation, 1994, Vol. 55-06B; ACS Symp. Ser.: Step-Growth Polymers for High-Performance Materials, 1996, Vol. 624, pp. 197-207; Macromolecules, 2000, Vol. 33 (10), pp. 3855-3859; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1999, Vol. 40 (2), pp. 620-621; Polym, Prepr. (Am. Chem.
20 Soc., Div. Polym. Chem.), 1999, Vol. 40 (2), pp. 577-78; Macromolecules, 1997, Vol. 30 (19), pp. 5970-5975; J. Polym. Sci, Part A: Polymer Chemistry, 1997, Vol. 35 (9), pp. 1743-1751; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1996, Vol. 37 (2), pp. 243-244; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1996, Vol. 37 (1), pp. 551-552; J. Polym. Sci., Part A: Polymer
25 Chemistry, 1996, Vol. 34 (3), pp. 397-402; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1995, Vol. 36 (2), pp. 140-141; Polym, Prepr. (Am. Chem. Soc., Div. Polym. Chem.), 1992, Vol. 33 (2), pp. 146-147; J. Appl. Polym. Sci., 1998, Vol. 68 (3), pp. 475-482). The adamantane-based compounds and monomers described by Reichert and Mathias are preferably used to form
30 polymers with adamantane molecules at the core of a thermoset. The

compounds disclosed by Reichert and Mathias in their studies, however, comprise only one isomer of the adamantane-based compound by design choice. Structure A shows this symmetrical *para*- isomer 1,3,5,7-tetrakis[4'-(phenylethynyl)phenyl]adamantane:



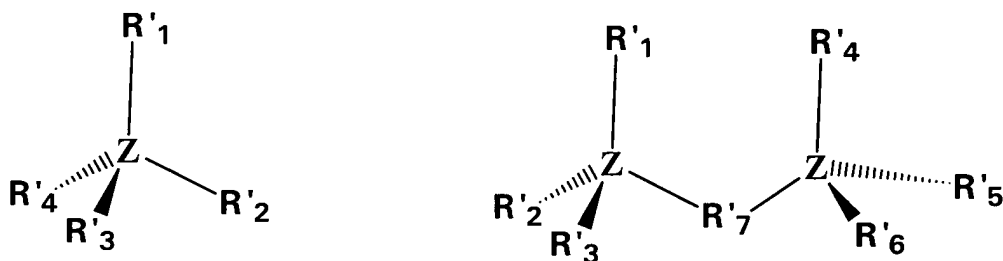
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Structure A

In other words, Reichert and Mathias in their individual and joint work contemplated a useful polymer comprising only one isomer form of the target adamantane-based monomer. A significant problem exists, however, when
10 forming and processing polymers from the single isomer form (symmetrical “all-*para*” isomer) 1,3,5,7-tetrakis[4'-(phenylethynyl)phenyl]adamantane of the adamantane-based monomer. According to the Reichert dissertation (*supra*) and *Macromolecules*, vol. 27, (pp. 7015-7034) (*supra*), the symmetrical all-*para* isomer 1,3,5,7-tetrakis[4'-(phenylethynyl)phenyl]adamantane “was found to be
15 soluble enough in chloroform that a ¹H NMR spectrum could be obtained. However, acquisition times were found to be impractical for obtaining a solution ¹³C NMR spectrum.” indicating that the all *para* isomer has low solubility. Thus, the Reichert symmetrical “all-*para*” isomer 1,3,5,7-tetrakis[4'-(phenylethynyl)phenyl]adamantane is insoluble in standard organic solvents and

therefore, would not be useful in any application requiring solubility or solvent-based processing, such as flow coating, spin coating, or dip coating.

In our commonly assigned pending patent application PCT/US01/22204 filed October 17, 2001 (claiming the benefit of our commonly assigned pending patent applications US Serial No. 09/545058 filed April 7, 2000; US Serial No. 09/618945 filed July 19, 2000; US Serial No. 09/897936 filed July 5, 2001; and US Serial No. 09/902924 filed July 10, 2001; and International Publication WO 01/78110 published October 18, 2001), we discovered a composition comprising an isomeric thermosetting monomer or dimer mixture, wherein the mixture comprises at least one monomer or dimer having the structure correspondingly



wherein Z is selected from cage compound and silicon atom; R'1, R'2, R'3, R'4, R'5, and R'6 are independently selected from aryl, branched aryl, and arylene ether; at least one of the aryl, the branched aryl, and the arylene ether has an ethynyl group; and R'7 is aryl or substituted aryl. We also disclose methods for formation of these thermosetting mixtures. This novel isomeric thermosetting monomer or dimer mixture is useful as a dielectric material in microelectronics applications and soluble in many solvents such as cyclohexanone. These desirable properties make this isomeric thermosetting monomer or dimer mixture ideal for film formation at thicknesses of about 0.1μm to about 1.0μm.

We filed a patent application Serial No. 60/_____ on even date herewith that claims a porous version of the preceding isomeric mixture.

Our International Patent Publication WO 01/78110 published October 18, 2001 teaches in its background section that methods for introducing nanosized voids include physical blending or chemical grafting of thermostable or thermolabile portions. This publication's invention is that nanosized voids may
5 be introduced into dielectric materials by using cage structures such as adamantane or diamantane to achieve low dielectric constant material and defines low dielectric constant materials as having a dielectric constant of less than 3.0. However, this publication does not report any dielectric constant for its examples.

10 International Patent Publication WO 00/31183 teaches in its background section that although known porous thermoplastic materials had acceptable dielectric constants, the pores tended to collapse during subsequent high temperature processing and thus, the art teaches away from adding porosity to the cage structure that introduced nanosized voids in International Patent
15 Publication WO 01/78110 published October 18, 2001. In addition, US Patents 5,776,990; 5,895,263; 6,107,357; and 6,342,454 and US Publication 2001/0040294 teach that although dielectric constants of 2.3-2.4 had been achieved at porosity levels less than about 20%, the pore content could not be further increased without comprising the small domain sizes and/or the non-
20 interconnectivity of the pore structure. Similarly, US Patents 6,271,273; 6,156,812; 6,171,687; and 6,172,128 teach that the amount of the thermally labile monomer unit is limited to amounts less than about 30% by volume because if more than about 30% by volume of the thermally labile monomer is used, the resulting dielectric material has cylindrical or lamellar domains, instead
25 of pores or voids, which lead to interconnected or collapsed structures upon removal, i.e., heating to degrade the thermally labile monomer units.

Although various methods are known in the art to lower the dielectric constant of a material, these methods have disadvantages. Thus, there is still a need in the semiconductor industry to a) provide improved compositions and
30 methods to lower the dielectric constant of dielectric layers; b) provide dielectric

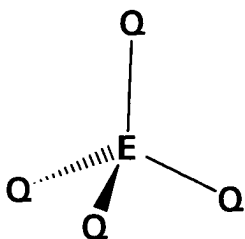
materials with improved properties, such as thermal stability, glass transition temperature (T_g), modulus, and hardness; c) produce thermosetting compounds and dielectric materials that are capable of being solvated and spun-on to a wafer or layered material; and d) provide materials with demonstrated
5 extendibility.

The present invention advantageously provides demonstrated extendibility so that semiconductor device manufacturers can use the present compositions for numerous generations of microchips. Also, the present invention provides for bonding of a porogen to a thermosetting component and thus, porogen
10 movement is minimized and the possibility of pore aggregation is reduced.

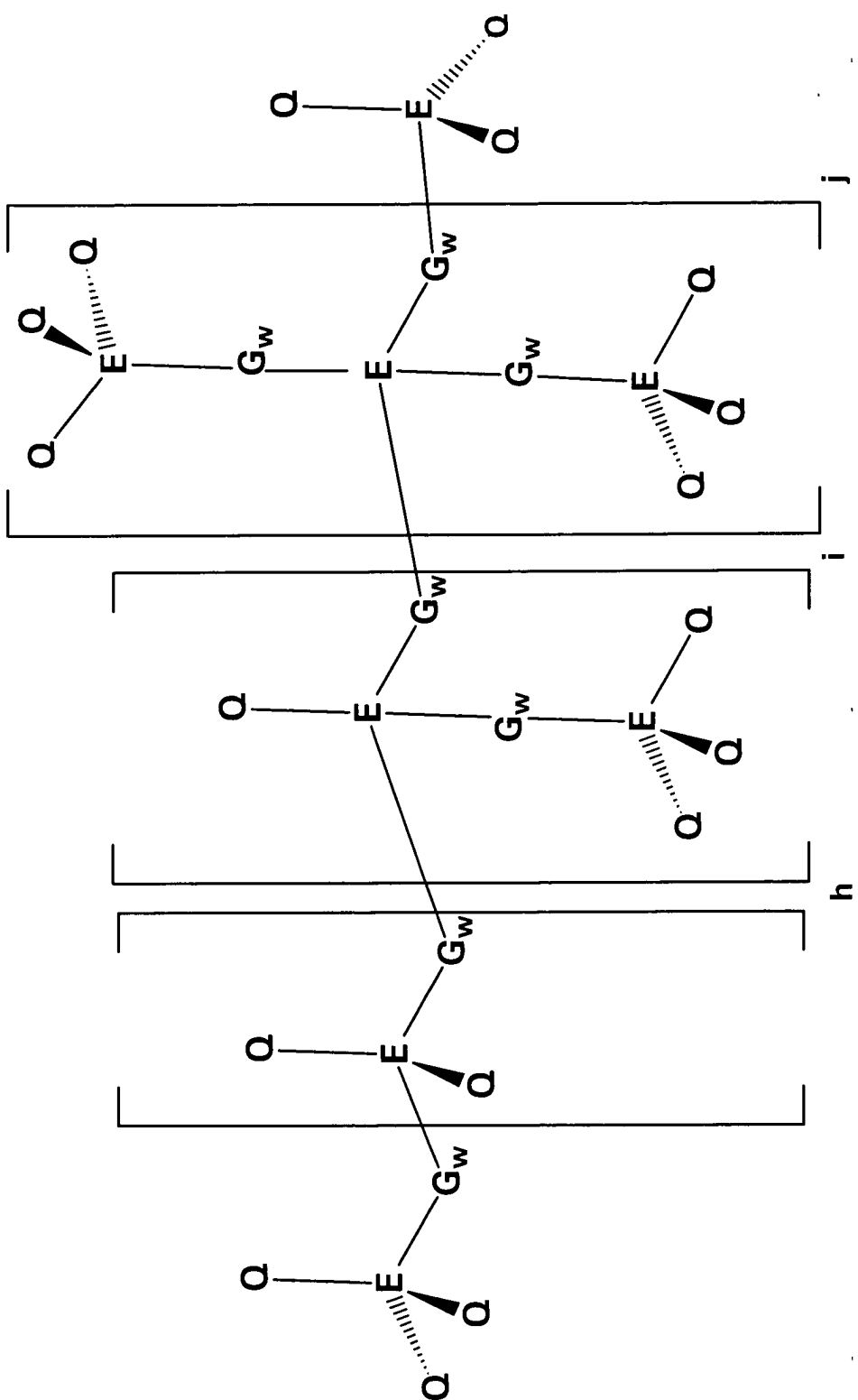
Summary of the Invention

In response to the need in the art and proceeding contrary to the wisdom in the art, we developed a composition comprising:

15 (a) thermosetting component comprising: (1) optionally monomer of Formula I



and (2) at least one oligomer or polymer of Formula II
20



where E is a cage compound; each of Q is the same or different and selected from aryl, branched aryl, and substituted aryl wherein said substituents include hydrogen, halogen, alkyl, aryl, substituted aryl, heteroaryl, aryl ether, alkenyl, alkynyl, alkoxy, hydroxyalkyl, hydroxyaryl, hydroxyalkenyl, hydroxyalkynyl, hydroxyl, or carboxyl; G is aryl or substituted aryl where substituents include halogen and alkyl; h is from 0 to 10; i is from 0 to 10; j is from 0 to 10; and w is 0 or 1;

(b) porogen that bonds to the thermosetting component (a).

We also discovered a method of lowering the dielectric constant of a composition comprising: (a) thermosetting component comprising: (1) optionally monomer of Formula I above and (2) at least one oligomer or polymer of Formula II above where E, Q, G, h, I, and j are defined as above; and

(b) adhesion promoter comprising compound having at least bifunctionality wherein the bifunctionality may be the same or different and the first functionality is capable of interacting with the thermosetting component (a) and the second functionality is capable of interacting with a substrate when the composition is applied to the substrate

comprising the steps of:

bonding porogen to the thermosetting component;

decomposing the bonded porogen; and

volatilizing the decomposed porogen whereby pores form in the composition.

Brief Description of the Drawings

Figures 1A through 1F illustrates how to make adamantane based compositions useful as the thermosetting component in the present compositions.

5 Figure 2 illustrates one method for making diamantane based compositions useful as the thermosetting component in the present compositions.

Figures 3A through 3F illustrate another method for making diamantane based compositions useful as the thermosetting component in the present
10 compositions.

Figures 4 through 11 illustrate reaction schemes for covalently bonding the thermosetting component to the porogen in the present compositions.

Figure 12 illustrates the reaction scheme of Inventive Example 1.

Figure 13 shows scanning electron microscopy pictures for the cross
15 section and surface of the film of Inventive Example 1.

Figure 14 illustrates the reaction scheme of Inventive Example 2.

Figure 15 shows scanning electron microscopy pictures for the cross section and surface of the film of Inventive Example 2.

Figure 16 illustrates the reaction scheme of Inventive Example 3.

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Detailed Description of the Invention

We filed a patent application 10/_____ on even date herewith that claims a composition of a thermosetting component and a porogen wherein the porogen does not have to bond to the thermosetting component.

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Thermosetting Component:

The phrases "cage structure", "cage molecule", and "cage compound" as used herein are intended to be used interchangeably and refer to a molecule
10 having at least eight atoms arranged such that at least one bridge covalently connects two or more atoms of a ring system. In other words, a cage structure, cage molecule, or cage compound comprises a plurality of rings formed by covalently bound atoms, wherein the structure, molecule, or compound defines a volume, such that a point located within the volume cannot leave the volume
15 without passing through the ring. The bridge and/or the ring system may comprise one or more heteroatoms, and may contain aromatic groups, partially cyclic or acyclic saturated hydrocarbon groups, or cyclic or acyclic unsaturated hydrocarbon groups. Further contemplated cage structures include fullerenes, and crown ethers having at least one bridge. For example, an adamantane or
20 diamantane is considered a cage structure, while a naphthalene or an aromatic spirocompound are not considered a cage structure under the scope of this definition, because a naphthalene or an aromatic spirocompound do not have one, or more than one bridge and thus, do not fall within the description of the cage compound above. Cage compounds are preferably adamantane and
25 diamantane and more preferably adamantane.

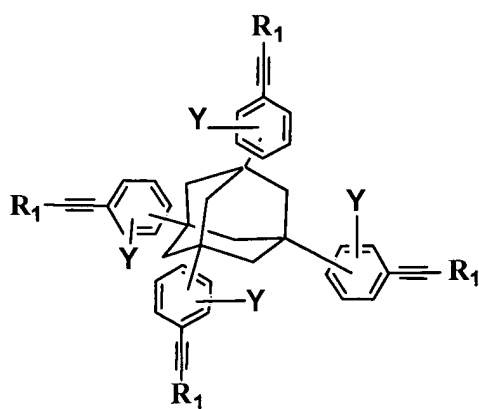
The phrase "bridgehead carbon" as used herein refers to any cage structure carbon bound to three other carbons. Thus, for example, adamantane has four bridgehead carbons while diamantane has eight bridgehead carbons.

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Preferred dielectric material is thermosetting component disclosed and claimed in our commonly assigned pending patent application 60/347195 filed January 8, 2002 and 60/_____ filed on even date herewith, which are incorporated herein by reference in their entirety.

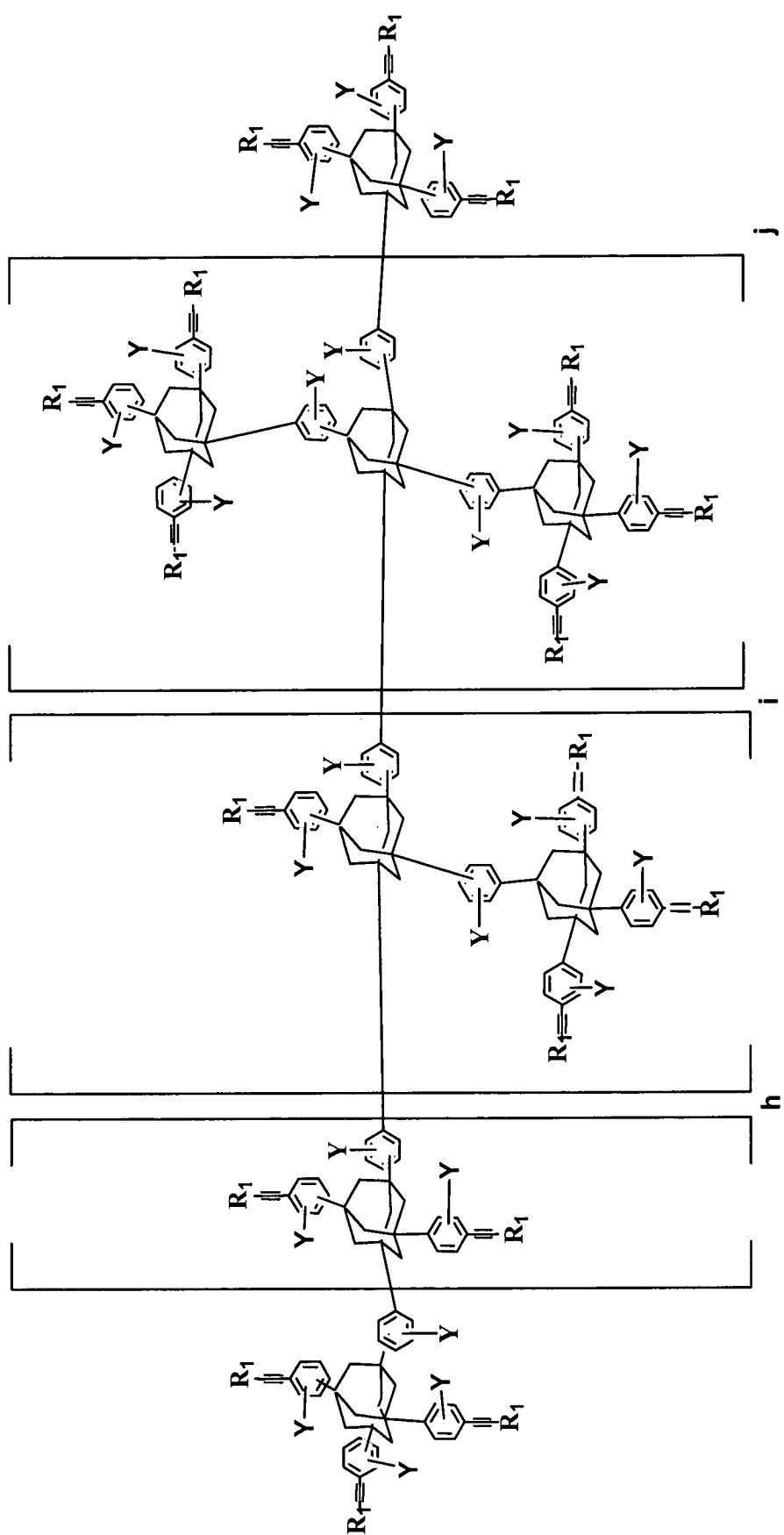
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Preferably, the thermosetting component (a) comprises: (1) adamantane monomer of Formula III

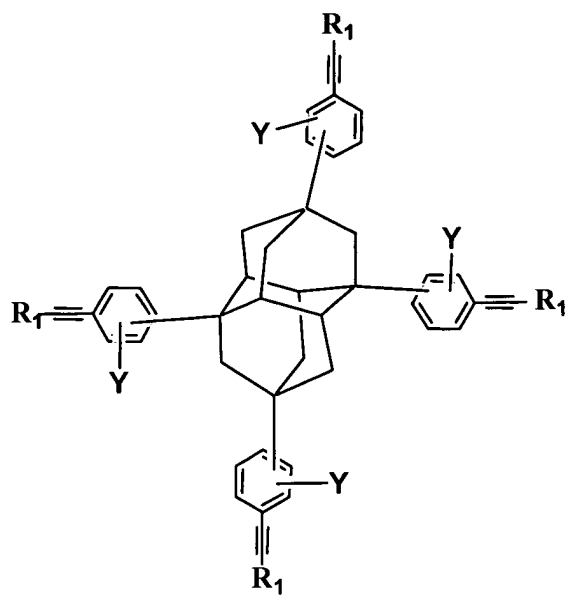


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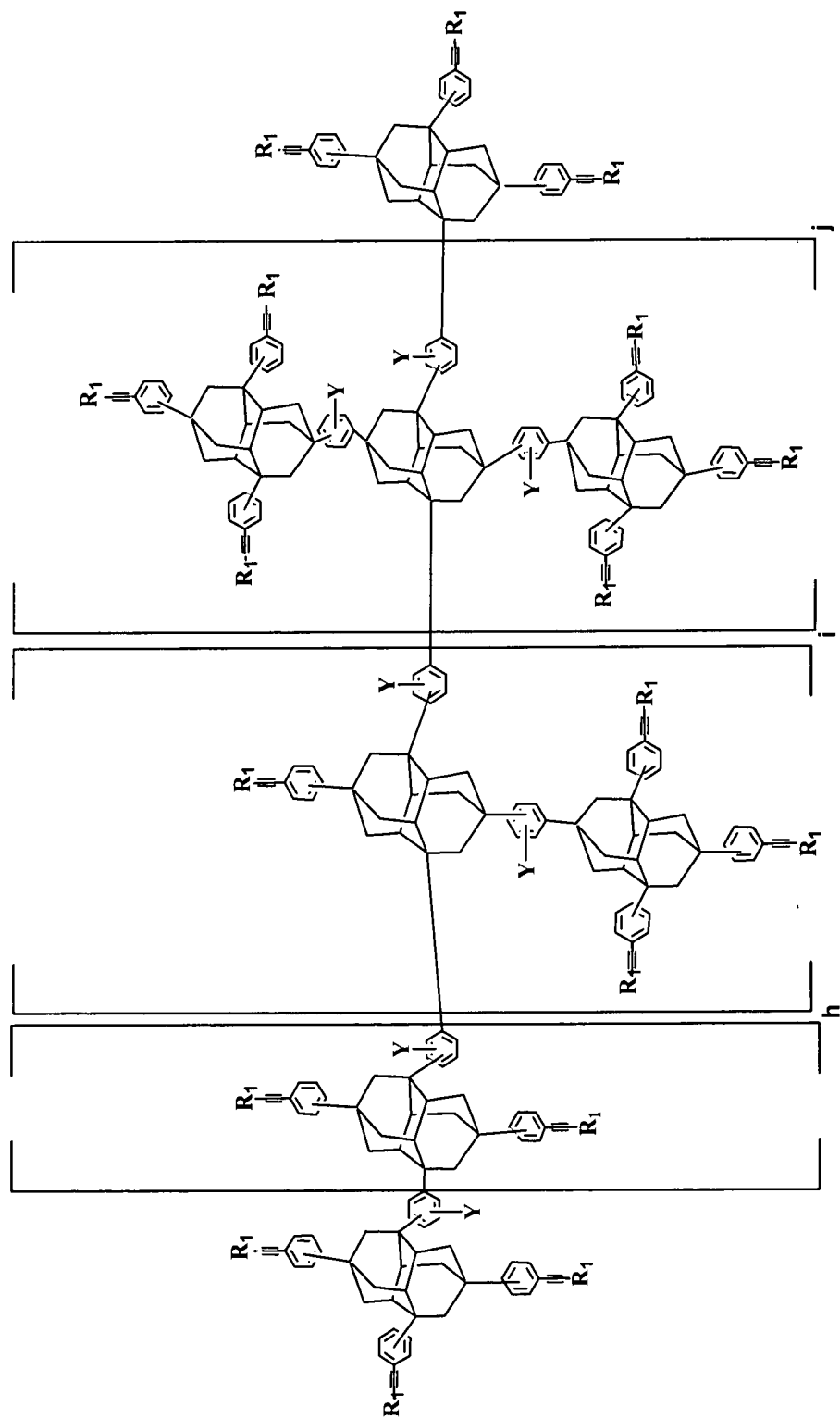
and (2) adamantane oligomer or polymer of Formula IV



or (1) diamantane monomer of Formula V



and (2) diamantane oligomer or polymer of Formula VI

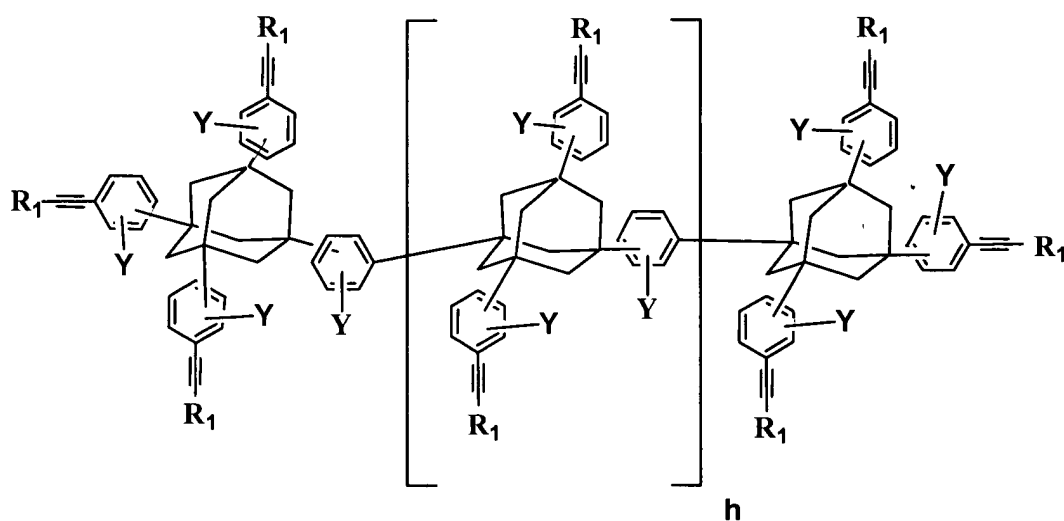


wherein h is from 0 to 10; i is from 0 to 10; j is from 0 to 10; each R₁ in Formulae III, IV, V, and VI is the same or different and selected from hydrogen, halogen, alkyl, aryl, substituted aryl, heteroaryl, aryl ether, alkenyl, alkynyl, alkoxyl, hydroxyalkyl, hydroxyaryl, hydroxyalkenyl, hydroxyalkynyl, hydroxyl, or carboxyl; and each Y in Formulae III, IV, V, and VI is the same or different and selected from hydrogen, alkyl, aryl, substituted aryl, or halogen.

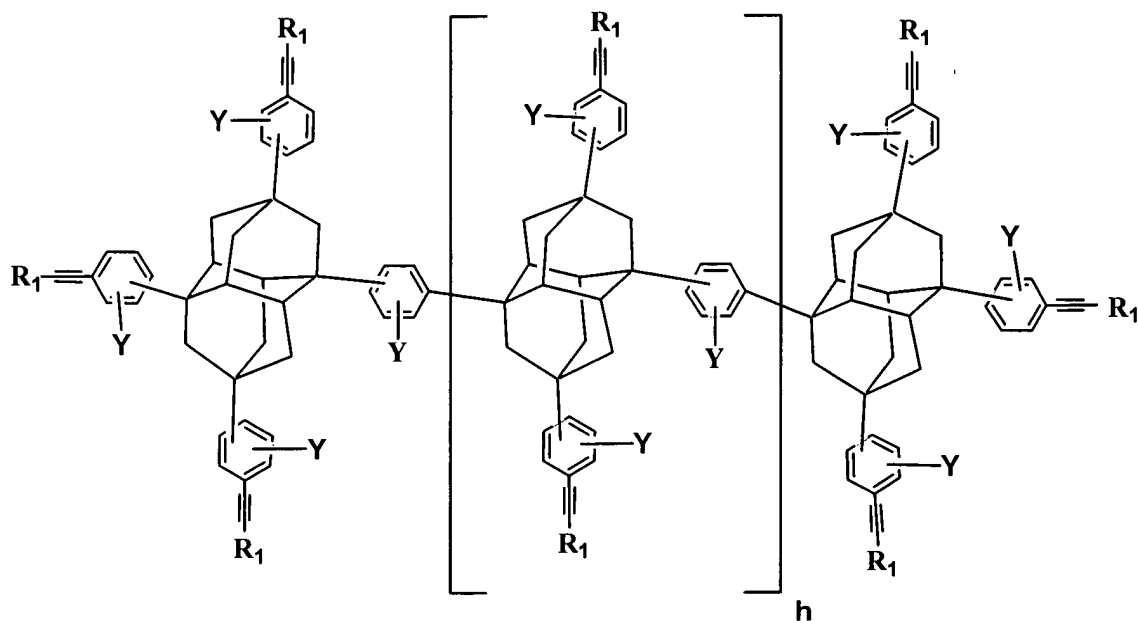
Formulae II, IV, and VI represent random or irregular structures in that any one of the units h, i, and j may or may not repeat numerous times before another unit is present. Thus, the sequence of units in Formulae II, IV, and VI above is random or irregular.

In the one embodiment, preferably the thermosetting component comprises adamantane monomer of Formula III above and at least one adamantane oligomer or polymer of Formula IV above where at least one of h, i, and j is at least 1. Preferably, the thermosetting component comprises diamantane monomer of Formula V above and at least one diamantane oligomer or polymer of Formula VI above where at least one of h, i, and j is at least 1.

Preferably, the thermosetting component comprises adamantane monomer of Formula III above and adamantane oligomer or polymer of Formula VII below.

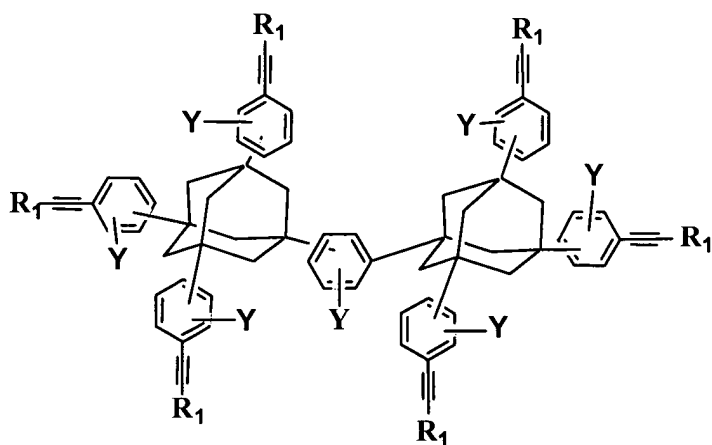


Preferably, the thermosetting component comprises diamantane monomer of Formula V above and diamantane oligomer or polymer of Formula VIII below.



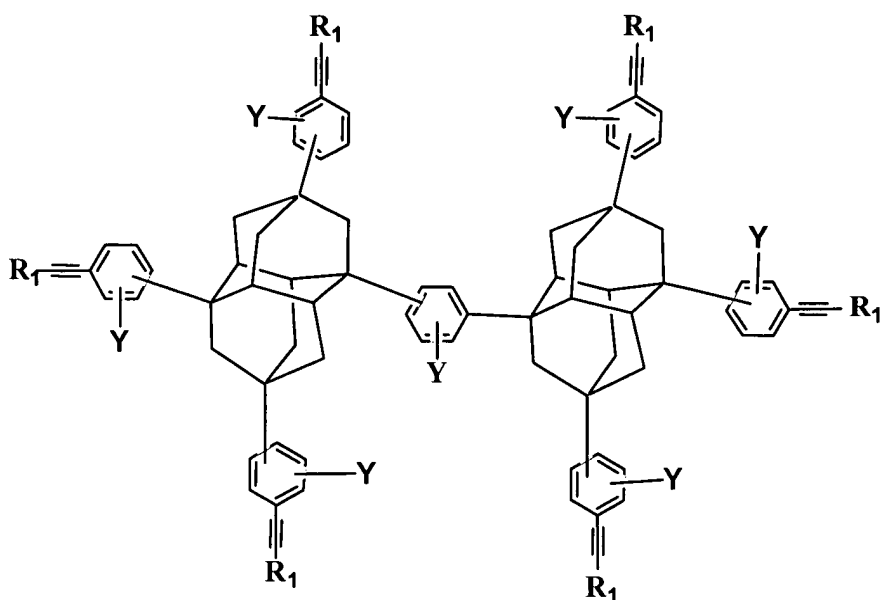
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Preferably, the thermosetting component comprises adamantane monomer of Formula III above and adamantane dimer of Formula IX below.



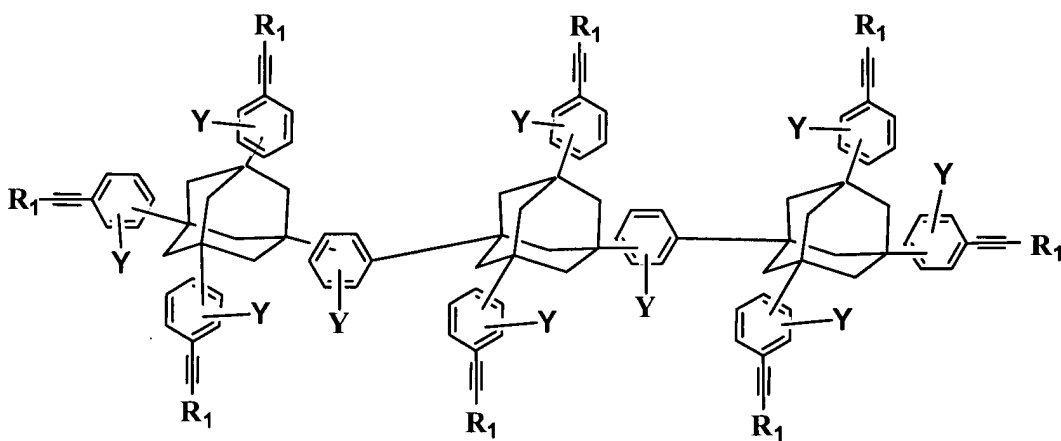
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Preferably, the thermosetting component comprises diamantane monomer of Formula V above and diamantane dimer of Formula X below.

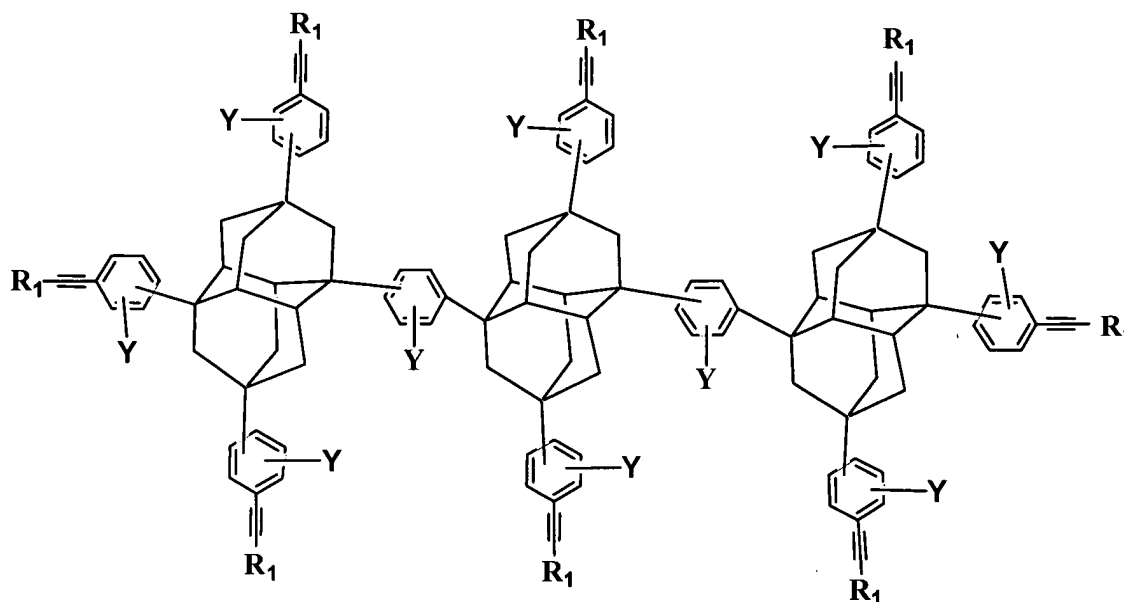


Preferably, the thermosetting component comprises adamantane monomer of Formula III above and adamantane trimer of Formula XI below.

5



Preferably, the thermosetting component comprises diamantane monomer of Formula V above and diamantane trimer of Formula XII below.



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Preferably, the thermosetting component comprises adamantane monomer of Formula III above, adamantane dimer of Formula IX above, and at least one adamantane oligomer or polymer of Formula IV above where at least one of h, i, and j is at least 1. Preferably, the thermosetting component comprises
 10 diamantane monomer of Formula IV above, diamantane dimer of Formula X above, and at least one diamantane oligomer or polymer of Formula VI above where at least one of h, i, and j is at least 1.

Preferably, the thermosetting component comprises adamantane monomer
 15 of Formula II above, adamantane dimer of Formula IX above, adamantane trimer of Formula XI above, and at least one adamantane oligomer or polymer of Formula IV above where at least one of i and j is at least 1. Preferably, the thermosetting component comprises diamantane monomer of Formula V above, diamantane dimer of Formula X above, diamantane trimer of Formula XII above,
 20 and at least one diamantane oligomer or polymer of Formula VI above where at least one of i and j is at least 1.

The thermosetting component comprises adamantane monomer of Formula III that is a tetrasubstituted adamantane or a diamantane monomer of Formula V that is a tetrasubstituted diamantane. The preferred monomer is the adamantane monomer of Formula III. The adamantane framework carries a substituted aryl radical in each of positions 1, 3, 5, and 7. The compound with the Formula IV is an oligomer or polymer, linked via unsubstituted and/or substituted aryl units, of the adamantane monomer of Formula III. The compound with the Formula VI is an oligomer or polymer, linked via unsubstituted and/or substituted aryl units, of the diamantane monomer of Formula V. Generally, h, i, and j are whole numbers from 0 to 10, preferably 0 to 5, and more preferably 0 to 2. The simplest adamantane oligomer is thus the dimer (h is 0, i is 0, and j is 0 in Formula IV) as shown in Formula IX above, in which two adamantane frameworks are linked via an unsubstituted or substituted aryl unit. The simplest diamantane oligomer is thus the dimer (h is 0, i is 0, and j is 0 in Formula VI) as shown in Formula X above, in which two diamantane frameworks are linked via an unsubstituted or substituted aryl unit.

In another embodiment, preferably the present thermosetting component comprises at least one adamantane oligomer or polymer of Formula IV above where h is from 0 to 10, i is from 0 to 10, and j is from 0 to 10. Preferably, the present thermosetting component comprises at least one diamantane oligomer or polymer of Formula VI above where h is from 0 to 10, i is from 0 to 10, and j is from 0 to 10.

Preferably, the present thermosetting component comprises at least one adamantane oligomer or polymer of Formula IV above where h is 0 or 1, i is 0, and j is 0. This adamantane structure is shown as Formula VII above.

Preferably, the present thermosetting component comprises at least one diamantane oligomer or polymer of Formula VI above where h is 0 or 1, i is 0, and j is 0. This diamantane structure is shown as Formula VIII above.

Preferably, the thermosetting component comprises at least one adamantane oligomer or polymer of Formula IV above where h is 0, i is 0, and j is 0. This adamantane dimer is shown as Formula IX above.

Preferably, the thermosetting component comprises at least one diamantane oligomer or polymer of Formula VI above where h is 0, i is 0, and j is 0. This diamantane dimer is shown as Formula X above.

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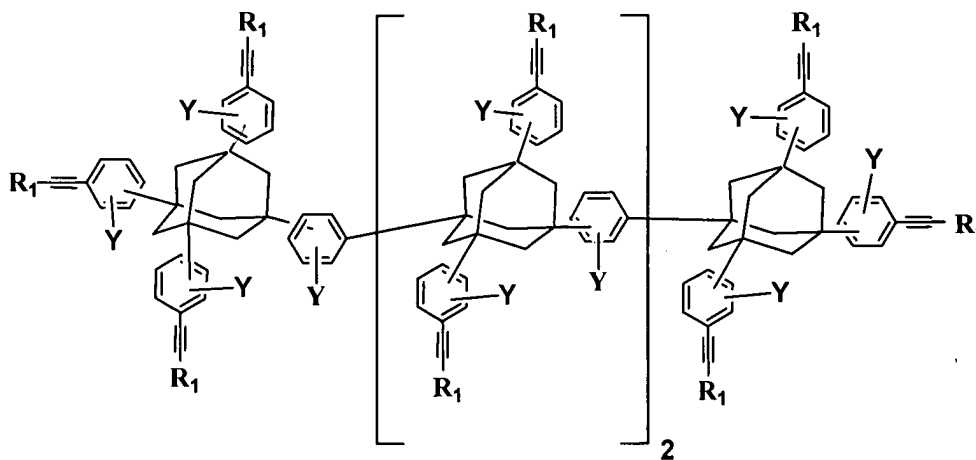
Preferably, the thermosetting component comprises at least one adamantane oligomer or polymer of Formula IV above where h is 1, i is 0, and j is 0. This adamantane trimer is as shown in Formula XI above.

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Preferably, the thermosetting component (a) comprises at least one diamantane oligomer or polymer of Formula VI above where h is 1, i is 0, and j is 0. This diamantane trimer is as shown in Formula XII above.

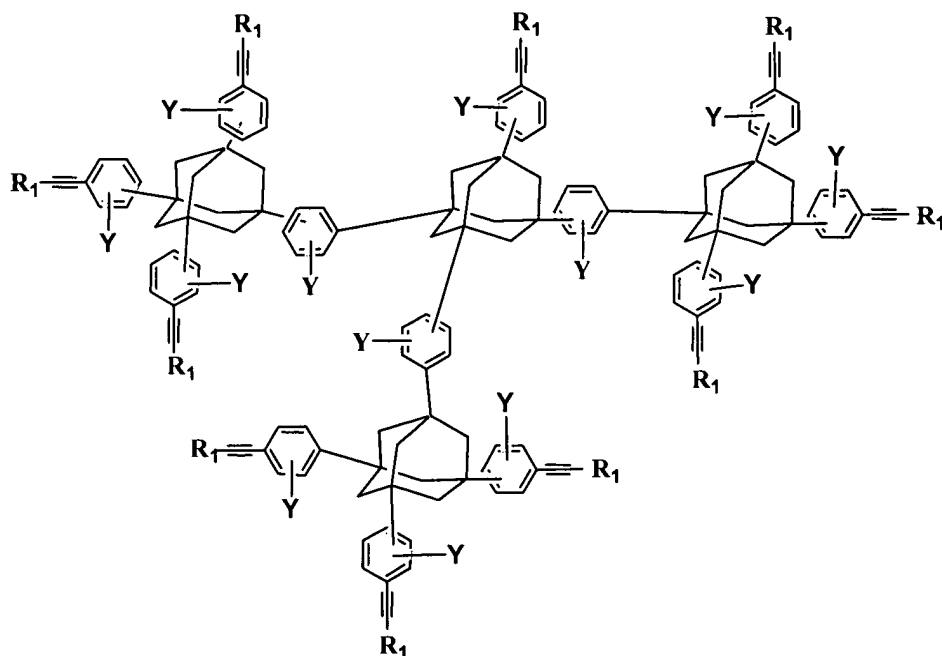
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Preferably, the thermosetting component comprises a mixture of at least one adamantane oligomer or polymer of Formula IV above where h is 2, i is 0, and j is 0 (linear oligomer or polymer) and h is 0, i is 1, and j is 0 (branched oligomer or polymer). Thus, this composition comprises a mixture of an adamantane linear tetramer as shown in Formula XIII below

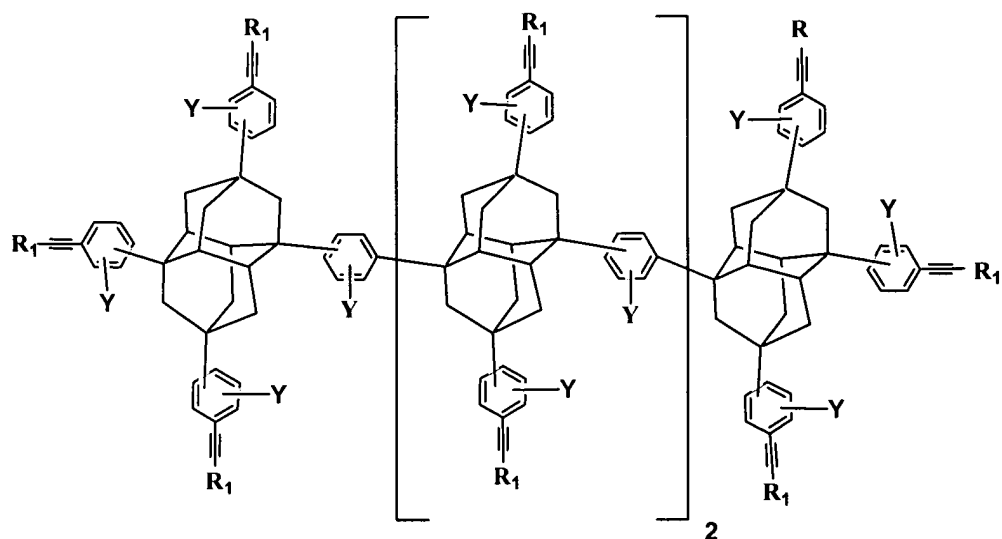


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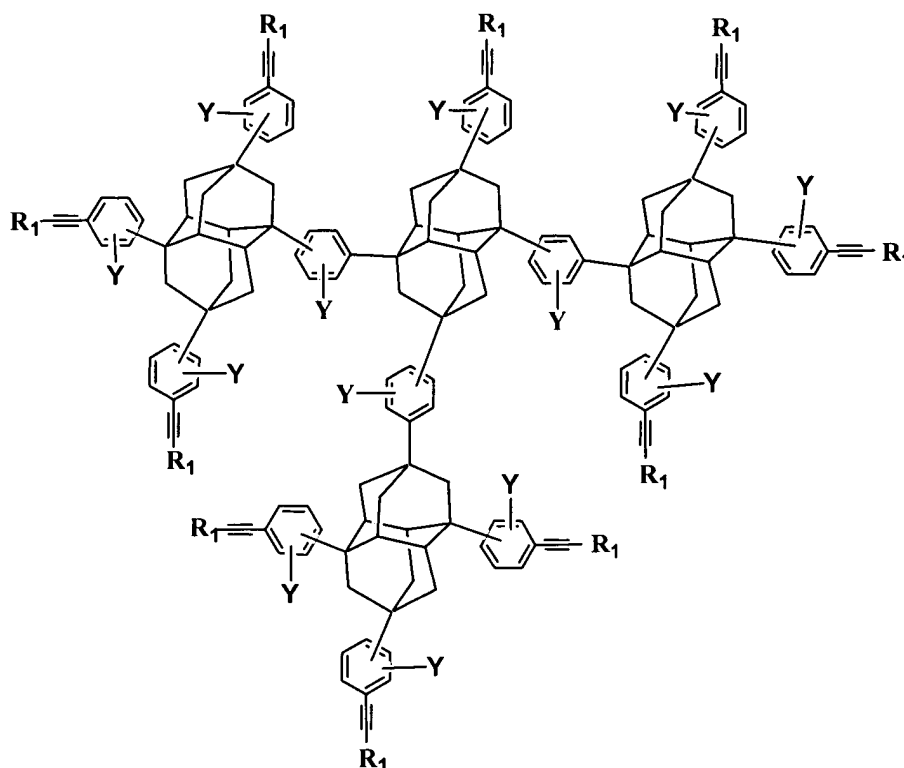
and adamantane branched tetramer as shown Formula XIV below



Preferably, the thermosetting component comprises at least one
 5 diamantane oligomer or polymer of Formula VI above where h is 2, i is 0, and j
 is 0 (linear oligomer or polymer) and h is 0, i is 1, and j is 0 (branched oligomer
 or polymer). Thus, the present composition comprises diamantane linear
 tetramer as shown in Formula XV below



and diamantane branched tetramer as shown Formula XVI below



5 Preferably, the thermosetting component comprises adamantane dimer of Formula IX above and adamantane trimer of Formula XI above. Preferably, the thermosetting component comprises diamantane dimer of Formula X above and diamantane trimer of Formula XII above.

10 Preferably, the thermosetting component comprises adamantane dimer of Formula IX above and at least one adamantane oligomer or polymer of Formula IV above where h is 0, i is at least 1, and j is 0. Preferably, the thermosetting component comprises diamantane dimer of Formula X above and at least one diamantane oligomer or polymer of Formula VI above where h is 0, i is at least
15 1, and j is 0.

In both embodiments, for Formulae I and II above, preferred Q groups include aryl and aryl substituted with alkenyl and alkynyl groups and more preferred Q groups include (phenylethynyl)phenyl,

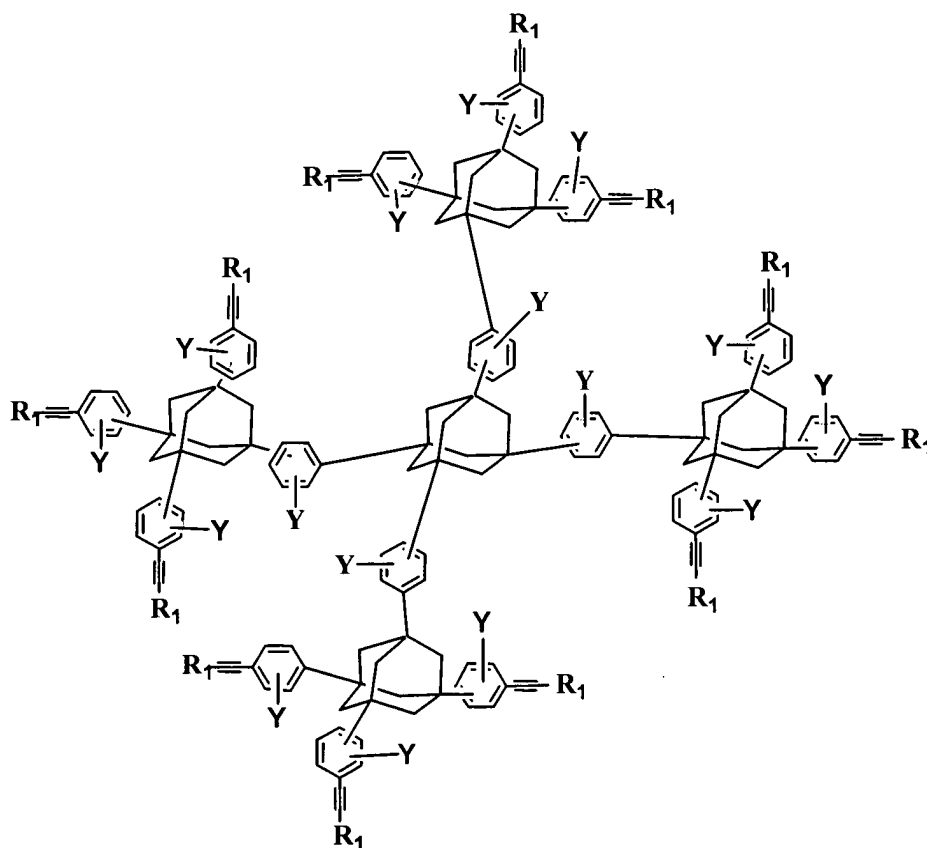
phenylethynyl(phenylethynyl)phenyl, and (phenylethynyl)phenylphenyl moiety. Preferred aryls for G include phenyl, biphenyl, and terphenyl. The more preferred G group is phenyl.

5 The individual radicals R_1 of the substituted ethynyl radical on the phenyl ring attached to the adamantane or diamantane ring of the type $R_1C\equiv C-$ are in each case the same or different in Formulae III, IV, V, VI, VII, VIII, IX, X, XI, XII, XIII, XIV, XV, and XVI above. R_1 is selected from hydrogen, halogen, alkyl, aryl, substituted aryl, heteroaryl, aryl ether, alkenyl, alkynyl, alkoxyl, hydroxyalkyl, hydroxyaryl, hydroxyalkenyl, hydroxyalkynyl, hydroxyl, or carboxyl. Each R_1 may be unbranched or branched and unsubstituted or substituted and the substituents may be unbranched or branched. It is preferred that the radicals alkyl, alkenyl, alkynyl, alkoxyl, hydroxyalkyl, hydroxyalkenyl, and hydroxyalkynyl contain from about 2 to about 10 carbon atoms and the radicals aryl, aryl ether, and hydroxyaryl contain from about 6 to about 18 carbon atoms. If R_1 stands for aryl, R_1 is preferably phenyl. Preferably, at least two of the $R_1C\equiv C$ groups on the phenyl groups are two different isomers. Examples of at least two different isomers include *meta*-, *para*-, and *ortho*- isomers. Preferably, the at least two different isomers are *meta*- and *para*- isomers. In the preferred monomer, 1,3,5,7-tetrakis[3'/4'-phenylethynyl]phenyl]adamantane (shown in Figure 1D), five isomers form: (1) *para*-, *para*-, *para*-, *para*-; (2) *para*-, *para*-, *para*-, *meta*-; (3) *para*-, *para*-, *meta*-, *meta*-; (4) *para*-, *meta*-, *meta*-, *meta*-; and (5) *meta*-, *meta*-, *meta*-, *meta*-.

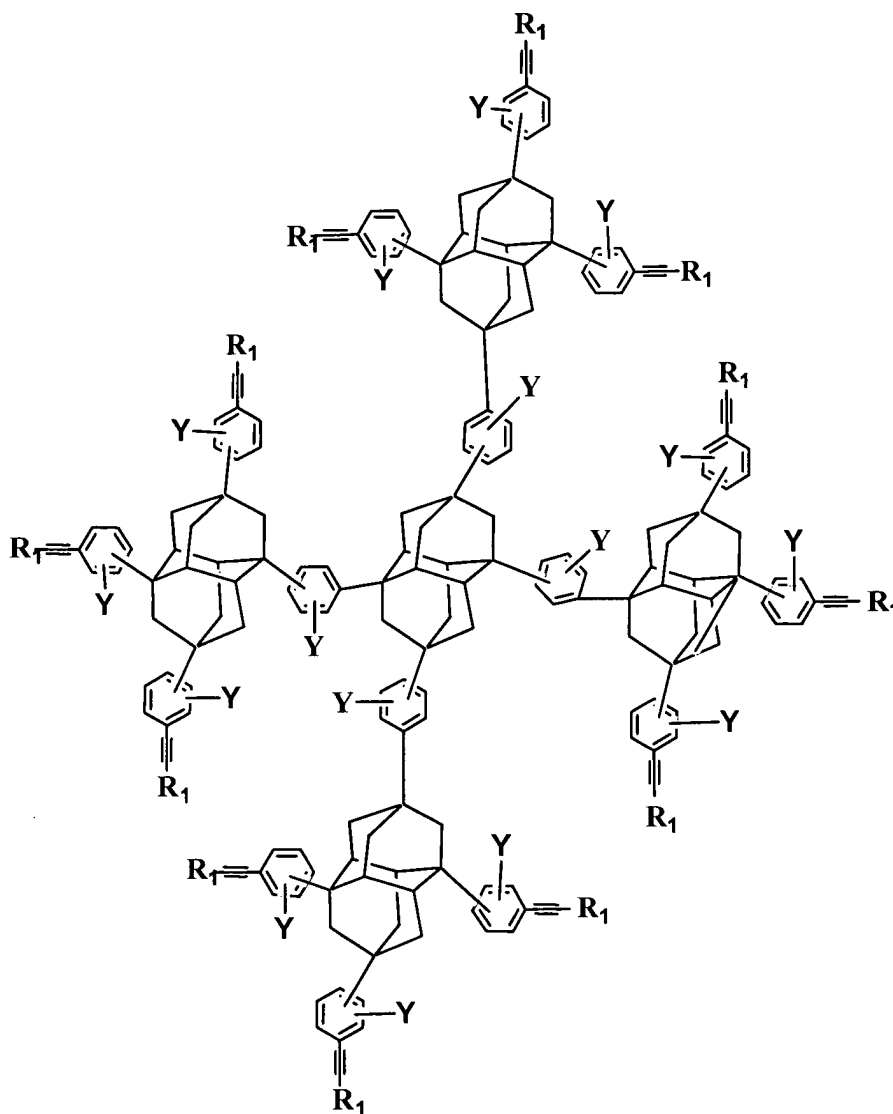
25 Each Y of the phenyl rings in the Formulae III, IV, V, VI, VII, VIII, IX, X, XI, XII, XIII, XIV, XV, and XVI above is in each case the same or different and selected from hydrogen, alkyl, aryl, substituted aryl, or halogen. When Y is aryl, examples of aryl groups include phenyl or biphenyl. Y is selected from preferably hydrogen, phenyl, and biphenyl and more preferably hydrogen. Preferably, at least one of the phenyl groups between two bridgehead carbons of adamantane or diamantane exists as at least two different isomers. Examples of at least two different isomers include *meta*-, *para*-, and *ortho*- isomers. Preferably, the at least two isomers are *meta*- and *para*- isomers. In the most preferred dimer 1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-

yl}benzene (shown in Figure 1F), 14 isomers form as follows. Preferably, the phenyl group located between the two bridgehead carbons of the adamantane exists as *meta*- and *para*- isomers. For each of the two preceding isomers, seven isomers of the $R_1C\equiv C$ groups on the phenyl groups exist as follows: (1) *para*-, *para*-, *para*-, *para*-, *para*-, *para*-; (2) *para*-, *para*-, *para*-, *para*-, *para*-, *meta*-; (3) *para*-, *para*-, *para*-, *para*-, *meta*-, *meta*-; (4) *para*-, *para*-, *para*-, *meta*-, *meta*-, *meta*-; (5) *para*-, *para*-, *meta*-, *meta*-, *meta*-, *meta*-; (6) *para*-, *meta*-, *meta*-, *meta*-, *meta*-, *meta*-; and (7) *meta*-, *meta*-, *meta*-, *meta*-, *meta*-, *meta*-.

In addition to the branched adamantane structure of Formula XIV above, it should be understood that Formula IV above when *h* is 0, *i* is 0, and *j* is 1 represents further branching as shown in Formula XVII below. It should be understood that branching may occur beyond that of the Formula XVII structure because further branching of the pending adamantane units of the Formula XVII structure may also occur.



In addition to the branched diamantane structure of Formula XVI above, it should be understood that Formula VI above when h is 0, i is 0, and j is 1 represents further branching as shown in Formula XVIII below. It should be understood that branching may occur beyond that of the Formula XVIII structure because further branching of the pending diamantane units of the Formula XVIII structure may also occur.



In thermosetting component, the monomer and oligomer or polymer contents are determined by the gel permeation chromatography techniques set forth below in the Analytical Test Methods section. The present composition

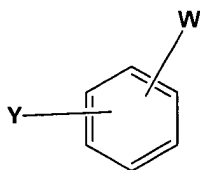
comprises the adamantane or diamantane monomer in a quantity of about 30 to about 70 area-%, more preferably about 40 to about 60 area-% and even more preferably about 45 to about 55 area-% and the oligomer or polymer in a quantity of about 70 to about 30 area-%, more preferably about 60 to about 40 area-%, and even more preferably about 55 to about 45 area-%. Most preferably, the present composition comprises the monomer (1) in a quantity of approximately 50 area-% and the oligomer or polymer (2) in a quantity of approximately 50 area-%.

The Analytical Test Methods section sets forth two Gel Permeation Chromatography Methods. Both provide similar results. One skilled in the art may elect to use the second method in that it yields additional detail on the dimer and trimer.

In general, the quantity ratio of the adamantane or diamantane monomer (1) to oligomer or polymer (2) can be set in a desired manner, e.g. by altering the molar ratio of the starting components during the preparation of the composition according to the invention, by adjusting reaction conditions, and by altering the ratio of nonsolvent to solvent during precipitation/isolation steps.

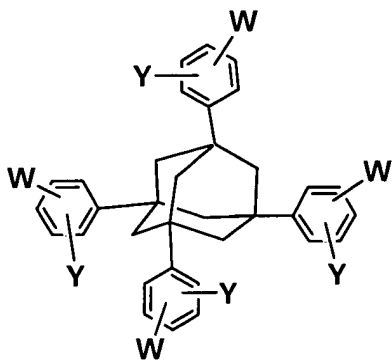
A preferred process for preparing the thermosetting component (a) comprises the following steps.

In step (A), adamantane or diamantane is reacted with halogeno benzene compound of Formula XIX



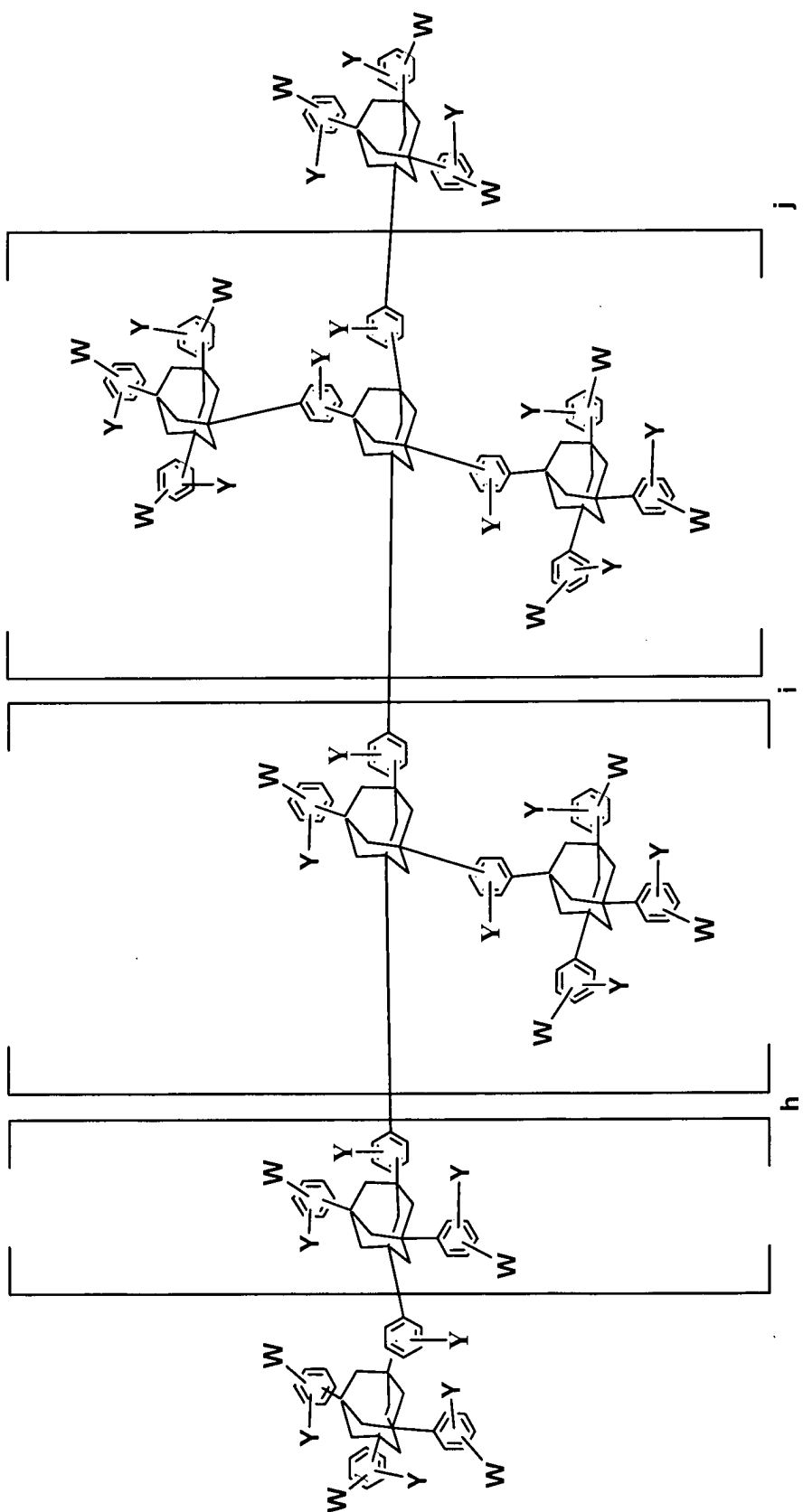
where W is halogen,

to form a mixture which if adamantane is used, comprises at least one monomer of Formula XX

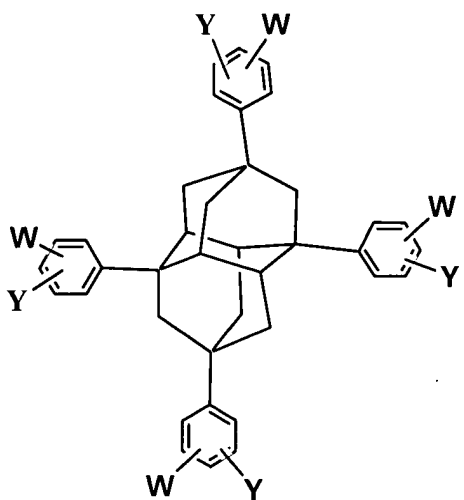


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and at least one oligomer or polymer of Formula XXI where h is from 0 to 10, i is from 0 to 10, and j is from 0 to 10

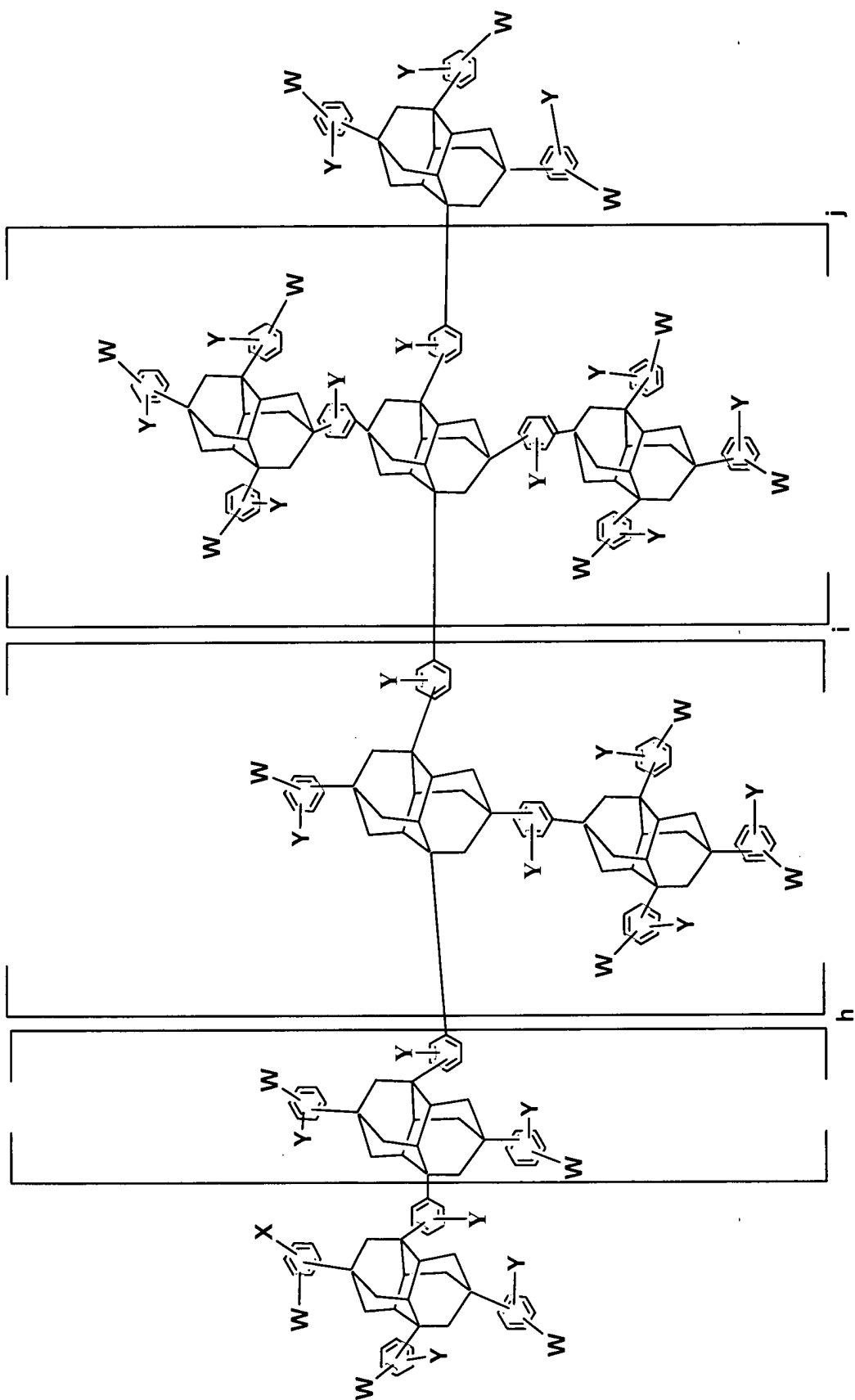


or if diamantane is used, comprises at least one monomer of Formula XXII



5

and at least one oligomer or polymer of Formula XXIII where h is from 0 to 10, i is from 0 to 10, and j is from 0 to 10



It should be understood to those skilled in the art that reaction may occur on diamantane at bridgehead carbons other than those indicated in Formulae XXII and XXIII above.

5 In step (B), the mixture resulting from step (A) is reacted with terminal alkyne of the formula $R_1C\equiv CH$. Preferably, the present process forms compositions of Formulae III and IV or V and VI above.

10 In step (A), adamantane or diamantane is reacted with halogeno benzene compound with the Formula XIX. In addition to the halogen radical W and the previously described radical Y, the halogeno benzene compound can also contain further substituents.

15 The halogeno benzene compound is preferably selected from bromobenzene, dibromobenzene, and iodobenzene. Bromobenzene and/or dibromobenzene are preferred, bromobenzene being even more preferred.

20 The reaction of adamantane or diamantane with the halogeno benzene compound (step (A)) takes place preferably through Friedel-Crafts reaction in the presence of a Lewis acid catalyst. Although all customary Lewis acid catalysts may be used, it is preferred that the Lewis acid catalyst contains at least one compound selected from aluminum(III) chloride ($AlCl_3$), aluminum(III) bromide ($AlBr_3$), and aluminum (III) iodide (AlI_3). Aluminum(III) chloride ($AlCl_3$) is most preferred. Despite the greater Lewis acidity of aluminum(III) bromide, its use is
25 generally less preferred, because it has a low sublimation point of only 90°C and is thus much more difficult to handle on an industrial scale than e.g. aluminum(III) chloride.

30 In a further preferred version, the Friedel-Crafts reaction is carried out in the presence of a second catalyst component. The second catalyst component preferably contains at least one compound selected from tertiary halogen alkane with 4 to 20 carbon atoms, tertiary alkanol with 4 to 20 carbon atoms, secondary and tertiary olefin with 4 to 20 carbon atoms and tertiary halogen alkyl aryl compound. In particular, the second catalyst component contains at
35 least one compound selected from 2-bromo-2-methylpropane (tert.-butyl bromide), 2-chloro-2-methylpropane (tert.-butyl chloride), 2-methyl-2-propanol (tert.-butyl alcohol), isobutene, 2-bromopropane, and tert.-butylbromobenzene,

with 2-bromo-2-methylpropane (tert.- butyl bromide) being most preferred. Overall, compounds whose alkyl groups include 5 or more carbon atoms are less suitable, as solid constituents precipitate out of the reaction solution at the end of the reaction.

5

It is most preferred that the Lewis acid catalyst is aluminum(III) chloride (AlCl_3) and the second catalyst component is 2-bromo-2-methylpropane (tert.-butyl bromide) or tert.-butylbromobenzene.

10 The preferable procedure for carrying out the Friedel-Crafts reaction is that adamantane or diamantane, halogeno benzene compound (e.g. bromobenzene), and Lewis acid catalyst (e.g. aluminium chloride) are mixed and heated at a temperature of 30°C to 50°C, preferably 35°C to 45°C and in particular 40°C. At temperatures lower than 30°C, the reaction is not completed,
15 i.e. a higher proportion of tri-substituted adamantane forms for example. In principle it is conceivable to use even higher temperatures than those given above (e.g. 60°C), but this leads in an undesirable manner to a higher proportion of non-halogenated aromatic material (e.g. benzene) in the reaction mixture of step (A). The second component of the catalyst system, say tert.-butyl bromide,
20 is then added to the above reaction solution generally over a period of 5 to 10 hours, preferably 6 to 7 hours and after the addition has ended, mixed into the reaction mixture in the temperature range named above customarily for a further 5 to 10 hours, preferably 7 hours.

25 Surprisingly, in addition to the monomeric tetraphenylated compound, e.g. 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane, oligomers or polymers thereof were also found in the mixture obtained after step (A). It was wholly unexpected that the quantity ratio of adamantane monomer of Formula XX to adamantane oligomer or polymer of Formula XXI or diamantane monomer of
30 Formula XXII to diamantane oligomer or polymer of Formula XXIII was controllable through the quantities of adamantane or diamantane, halogeno benzene compound (e.g. bromobenzene), and second catalyst component (e.g. tert.-butyl bromide) used. The molar ratio of adamantane or diamantane to halogeno benzene compound to second catalyst component in the reaction
35 mixture of step (A) is preferably 1:(5-15):(2-10) and even more preferably 1:(8-12):(4-8).

In the compounds with the Formulae XX, XXI, XXII, and XXIII, the position of the halogen substituent W is undefined. Preferably, the mixtures comprise *meta*- and *para*-isomers which, unlike all *para*-isomers, advantageously produce improved solubility and good film properties. In the reaction mixture of step (A), in addition to monomers and oligomers or polymers, starting components and by-products, such as not wholly phenylated adamantanes, can also occur.

The mixture resulting from step (A) is optionally worked up using methods known to those skilled in the art. For example, it may be necessary to remove non-reacted halogen phenyl compound, say bromobenzene, from the mixture in order to obtain a product, usable for further reaction, with a high proportion of compounds of Formulae XX, XXI, XXII, and XXIII. Any solvent or solvent mixture which is miscible with the halogeno benzene compound, say bromobenzene, and is suitable for the precipitation of the compounds of Formulae XX, XXI, XXII, and XXIII may be used for the isolation of such a product. It is preferred to introduce the mixture resulting from step (A) into a nonpolar solvent or solvent mixture, e.g. by dropping in, with preference being given to the use of aliphatic hydrocarbons with 7 to 20 carbon atoms or mixtures thereof and in particular at least one component selected from heptane fraction (boiling point 93-99°C), octane fraction (boiling point 98-110°C) and alkane mixture currently commercially available from Honeywell International Inc. under the tradename Spezial Benzin 80-110°C (petroleum ether with boiling point of 80-110°C). Spezial Benzin 80-110°C (petroleum ether with boiling point of 80-110°C) is most preferred. The weight ratio of organic mixture to nonpolar solvent is preferably about 1:2 to about 1:20, more preferably about 1:5 to about 1:13, and even more preferably about 1:7 to about 1:11. Alternatively, a polar solvent or solvent mixture (e.g. methanol or ethanol) can be used for the working-up of the mixture obtained after step (A), but it is less preferred, as the product mixture then precipitates out as a rubbery composition.

We have found that the peak ratio of monomer resulting from step (A) above to its dimer and trimer and oligomer in the reaction mixture shifts dramatically if the step (A) mixture is precipitated into certain solvents. This discovery advantageously allows one skilled in the art to adjust process conditions in order to achieve a targeted ratio of monomer to dimer and trimer

and oligomer. To reduce this ratio, preferably, a solvent is used in which the monomer and oligomer or polymer have different solubilities.

Preferred solvents for achieving this monomer to dimer and trimer ratio shift include Spezial Benzin 80-110°C (petroleum ether with boiling point of 80°C-110°C), ligroine (boiling point 90-110°C), and heptane (boiling point 98°C). The more preferred solvent is Spezial Benzin. More specifically, to achieve a shift from about 3:1 monomer:dimer+trimer+oligomer to about 1:1, the step (A) mixture is precipitated into Spezial Benzin or to attain a shift from about 3:1 monomer:dimer+trimer+oligomer to about 1.7-2.0:1.0, the step (A) reaction mixture is precipitated into ligroine and heptane. We know that these substantial changes in peak distribution at precipitation are explained by the loss of monomer in the precipitation filtrates: 2/3 loss in Spezial Benzin and $\geq 1/3$ loss in ligroine and heptane, which correspond to monomer yield losses of 50 and 25-33%. In order for the ratio monomer:dimer+trimer+oligomer 3:1 to remain unchanged, the step (A) reaction mixture is precipitated into methanol where no yield losses are observed. This is corroborated by determination of yield losses of the filtrates and GPC analysis of the filtrates.

Like the synthesis described by Ortiz, the Friedel-Crafts reaction which is carried out according to a preferred version in step (A) of the present process starts direct from adamantane which is coupled with the halogeno benzene compound. Compared with previous syntheses of e.g. 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane by Reichert et al., the present process is particularly advantageous because it is no longer necessary to produce tetrabrominated adamantanes first, which saves a reaction step. Also, less unwanted benzene forms.

It is known to those skilled in the art that the halogen radical W in the compounds of Formulae XX, XXI, XXII, and XXIII above can also be introduced, apart from a direct reaction of adamantane with halogen phenyl compound (e.g. with the help of a Friedel-Crafts reaction), by a multi-stage synthesis, for example, by coupling adamantane with a phenyl compound (i.e. without halogen

radical W) followed by introduction of the radical W say through addition with W_2 (e.g. Br_2) although this is not preferred.

5 In step (B) of the preferred process, the (optionally worked-up) mixture obtained after step (A) is reacted with terminal alkyne of the formula $R_1C\equiv CH$ where R_1 is as previously defined.

10 In the formula $R_1C\equiv CH$, R_1 is identical with the previously described radical R_1 of the adamantane product of Formulae III and IV and the diamantane product of Formulae V and VI. Accordingly it is most preferred to use ethynyl benzene (phenylacetylene) as terminal alkyne for the reaction in step (B).

15 In order, in step (B), to couple the terminal alkyne to the halogeno benzene radicals located at the adamantane system, all conventional coupling methods suitable for this purpose may be used, as described for example in Diederich, F., and Stang, P.J., (Eds.) "Metal-Catalyzed Cross-Coupling Reactions", Wiley-VCH 1998 and March, J., "Advanced Organic Chemistry", 4th Edition, John Wiley & Sons 1992, pages 717/718.

20 When Y on the phenyl groups is attached to two cage structure bridgehead carbons in Formula XXI above or in Formula XXIII above, Y may react with phenylacetylene to generate terminal alkyne groups.

25 In a preferred version of the process according to the invention, the reaction of the (optionally worked-up) mixture obtained after step (A) with terminal alkyne is carried out in the presence of a catalyst system as used in the so-called Sonogashira coupling (cf. Sonogashira; Tohda; Hagihara; Tetrahedron Lett. 1975, page 4467). It is even more preferred to use a catalyst system which in each case contains at least one palladium-triarylphosphine complex with the formula $[Ar_3P]_2PdX_2$ (where Ar = aryl and X = halogen), a copper halide (e.g. CuI), a base (e.g. a trialkylamine), a triarylphosphine and a co-solvent. According to the invention, this preferred catalyst system can equally well consist of the named components. The co-solvent preferably contains at least one component selected from toluene, xylene, chlorobenzene, N,N-dimethylformamide and 1-methyl-2-pyrrolidone (N-methylpyrrolidone (NMP)). A catalyst system which contains the components bis-(triphenylphosphine)palladium(II)dichloride (i.e. $[Ph_3P]_2PdCl_2$), triphenylphosphine

(i.e. $[\text{Ph}_3\text{P}]$), copper(I)-iodide, triethylamine and toluene as co-solvent is most preferred.

The preferred procedure for the reaction of the mixture obtained from step (A) (and optionally worked-up) with terminal alkyne is that the mixture is first mixed with the base (e.g. triethylamine) and the co-solvent (e.g. toluene) and this mixture is stirred for some minutes at room temperature. The palladium-triphenylphosphine complex (e.g. $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$), triphenylphosphine (PPh_3) and copper halide (e.g. copper(I)-iodide) are then added, and this mixture is heated in a temperature range of 50°C to 90°C (more preferably 80°C to 85°C). Terminal alkyne is then added in the named temperature range within 1 to 20 hours (more preferably 3 hours). After the ending of the addition, the mixture is heated for at least 5 to 20 hours (more preferably 12 hours) at a temperature of 75°C to 85°C (more preferably 80°C). Solvent is then added to the reaction solution and distilled off under reduced pressure. Preferably, after filtration, the reaction solution is then cooled to a temperature of 20°C to 30°C (more preferably 25°C). Finally, the reaction mixture of step (B), in particular for the removal of metal traces (e.g. Pd), is worked up with conventional methods which are known to those skilled in the art.

20

The peak ratio of monomer resulting from step (B) above to its dimer and trimer and oligomer in the reaction mixture shifts if the step (B) mixture is precipitated into certain solvents.

Surprisingly, it transpired that the reaction sequence starting direct from adamantane leads to an oligomeric or polymeric content in the reaction product of step (A) which can be controlled via the use ratio of adamantane, halogeno benzene compound and the second catalyst component, say tert.-butyl bromide. In corresponding manner, the benzene content in the reaction mixture of step (A) is also successfully regulated via this use ratio, which, because of the toxicity of benzene in industrial-scale syntheses, is of great importance. The oligomeric or polymeric content permits the same secondary chemistry as the

30

monomer (e.g. 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane, i.e. the oligomer or polymer is just as accessible as the monomer for the reaction with the terminal alkyne in step (B)).

5

Adhesion Promoter:

The phrase "adhesion promoter" as used herein means any component that when added to thermosetting component, improves the adhesion thereof to
10 substrates compared with thermosetting component alone.

The phrase "compound having at least bifunctionality" as used herein means any compound having at least two functional groups capable of interacting or reacting, or forming bonds as follows. The functional groups may
15 react in numerous ways including addition reactions, nucleophilic and electrophilic substitutions or eliminations, radical reactions, etc. Further alternative reactions may also include the formation of non-covalent bonds, such as Van der Waals, electrostatic bonds, ionic bonds, and hydrogen bonds.

20 Adhesion promoter is disclosed in our commonly assigned pending patent application 60/350187 filed January 15, 2002 which is incorporated herein by reference in its entirety.

In the adhesion promoter, preferably at least one of the first functionality
25 and the second functionality is selected from Si containing groups; N containing groups; C bonded to O containing groups; hydroxyl groups; and C double bonded to C containing groups. Preferably, the Si containing groups are selected from Si-H, Si-O, and Si-N; the N containing groups are selected from such as C-NH₂ or other secondary and tertiary amines, imines, amides, and
30 imides; the C bonded to O containing groups are selected from =CO, carbonyl

groups such as ketones and aldehydes, esters, -COOH, alkoxyls having 1 to 5 carbon atoms, ethers, glycidyl ethers; and epoxies; the hydroxyl group is phenol; and the C double bonded to C containing groups are selected from allyl and vinyl groups. For semiconductor applications, the more preferred functional groups include the Si containing groups; C bonded to O containing groups; hydroxyl groups; and vinyl groups.

An example of a preferred adhesion promoter having Si containing groups is silanes of the Formula XXIV: $(R_2)_k(R_3)_lSi(R_4)_m(R_5)_n$ wherein R_2 , R_3 , R_4 , and R_5 each independently represents hydrogen, hydroxyl, unsaturated or saturated alkyl, substituted or unsubstituted alkyl where the substituent is amino or epoxy, saturated or unsaturated alkoxyl, unsaturated or saturated carboxylic acid radical, or aryl; at least two of R_2 , R_3 , R_4 , and R_5 represent hydrogen, hydroxyl, saturated or unsaturated alkoxyl, unsaturated alkyl, or unsaturated carboxylic acid radical; and $k+l+m+n \leq 4$. Examples include vinylsilanes such as $H_2C=CHSi(CH_3)_2H$ and $H_2C=CHSi(R_6)_3$ where R_6 is CH_3O , C_2H_5O , AcO , $H_2C=CH$, or $H_2C=C(CH_3)O-$, or vinylphenylmethylsilane; allylsilanes of the formula $H_2C=CHCH_2-Si(OC_2H_5)_3$ and $H_2C=CHCH_2-Si(H)(OCH_3)_2$; glycidoxypropylsilanes such as (3-glycidoxypropyl)methyldiethoxysilane and (3-glycidoxypropyl)trimethoxysilane; methacryloxypropylsilanes of the formula $H_2C=(CH_3)COO(CH_2)_3-Si(OR_7)_3$ where R_7 is an alkyl, preferably methyl or ethyl; aminopropylsilane derivatives including $H_2N(CH_2)_3Si(OCH_2CH_3)_3$, $H_2N(CH_2)_3Si(OH)_3$, or $H_2N(CH_2)_3OC(CH_3)_2CH=CHSi(OCH_3)_3$. The aforementioned silanes are commercially available from Gelest.

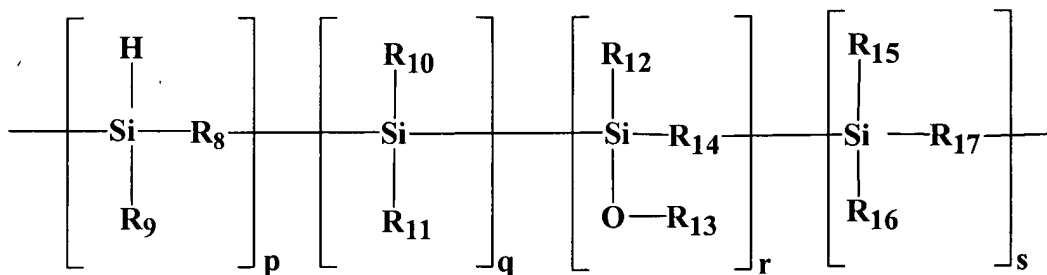
An example of a preferred adhesion promoter having C bonded to O containing groups is glycidyl ethers including but not limited to 1,1,1-tris-(hydroxyphenyl)ethane tri-glycidyl ether which is commercially available from TriQuest.

An example of a preferred adhesion promoter having C bonded to O containing groups is esters of unsaturated carboxylic acids containing at least

one carboxylic acid group. Examples include trifunctional methacrylate ester, trifunctional acrylate ester, trimethylolpropane triacrylate, dipentaerythritol pentaacrylate, and glycidyl methacrylate. The foregoing are all commercially available from Sartomer.

5 An example of a preferred adhesion promoter having vinyl groups is vinyl cyclic pyridine oligomers or polymers wherein the cyclic group is pyridine, aromatic, or heteroaromatic. Useful examples include but not limited to 2-vinylpyridine and 4-vinylpyridine, commercially available from Reilly; vinyl aromatics; and vinyl heteroaromatics including but not limited to vinyl quinoline,
10 vinyl carbazole, vinyl imidazole, and vinyl oxazole.

An example of a preferred adhesion promoter having Si containing groups is the polycarbosilane disclosed in commonly assigned copending allowed US patent application Serial 09/471299 filed December 23, 1999 incorporated
15 herein by reference in its entirety. The polycarbosilane is of the Formula XXV:



in which R₈, R₁₄, and R₁₇ each independently represents substituted or
20 unsubstituted alkylene, cycloalkylene, vinylene, allylene, or arylene; R₉, R₁₀, R₁₁, R₁₂, R₁₃, R₁₅, and R₁₆ each independently represents hydrogen atom or organo group comprising alkyl, alkylene, vinyl, cycloalkyl, allyl, or aryl and may be linear or branched; R₁₃ represents organosilicon, silanyl, siloxyl, or organo group; and
25 p, q, r, and s satisfy the conditions of $[4 \leq p + q + r + s \leq 100,000]$, and q and r and s may collectively or independently be zero. The organo groups may contain up to 18 carbon atoms but generally contain from about 1 to about 10 carbon atoms. Useful alkyl groups include -CH₂- and -(CH₂)_t- where t > 1.

Preferred polycarbosilanes of the present invention include dihydrido polycarbosilanes in which R_8 is a substituted or unsubstituted alkylene or phenyl, R_9 group is a hydrogen atom and there are no appendent radicals in the polycarbosilane chain; that is, q , r , and s are all zero. Another preferred group of polycarbosilanes are those in which the R_9 , R_{10} , R_{11} , R_{12} , R_{15} , and R_{16} groups of Formula XXV are substituted or unsubstituted alkenyl groups having from 2 to 10 carbon atoms. The alkenyl group may be ethenyl, propenyl, allyl, butenyl or any other unsaturated organic backbone radical having up to 10 carbon atoms. The alkenyl group may be dienyl in nature and includes unsaturated alkenyl radicals appended or substituted on an otherwise alkyl or unsaturated organic polymer backbone. Examples of these preferred polycarbosilanes include dihydrido or alkenyl substituted polycarbosilanes such as polydihydridocarbosilane, polyallylhypididocarbosilane and random copolymers of polydihydridocarbosilane and polyallylhypididocarbosilane.

In the more preferred polycarbosilanes, the R_9 group of Formula XXV is a hydrogen atom and R_8 is methylene and the appendent radicals q , r , and s are zero. Other preferred polycarbosilane compounds of the invention are polycarbosilanes of Formula XXV in which R_9 and R_{15} are hydrogen, R_8 and R_{17} are methylene, and R_{16} is an alkenyl, and appendent radicals q and r are zero. The polycarbosilanes may be prepared from well known prior art processes or provided by manufacturers of polycarbosilane compositions. In the most preferred polycarbosilanes, the R_9 group of Formula XXV is a hydrogen atom; R_8 is $-CH_2-$; q , r , and s are zero and p is from 5 to 25. These most preferred polycarbosilanes may be obtained from Starfire Systems, Inc. Specific examples of these most preferred polycarbosilanes follow:

Polycarbosilane	Weight Average Molecular Weight (Mw)	Polydispersity	Peak Molecular Weight (Mp)
1	400-1,400	2-2.5	330-500
2	330	1.14	320

3 (with 10% allyl groups)	10,000-14,000	10.4-16	1160
4 (with 75% allyl groups)	2,400	3.7	410

As can be observed in Formula XXV, the polycarbosilanes utilized in the subject invention may contain oxidized radicals in the form of siloxyl groups when $r > 0$. Accordingly, R_{13} represents organosilicon, silanyl, siloxyl, or organo group when $r > 0$. It is to be appreciated that the oxidized versions of the polycarbosilanes ($r > 0$) operate very effectively in, and are well within the purview of the present invention. As is equally apparent, r can be zero independently of p , q , and s the only conditions being that the radicals p , q , r , and s of the Formula XXV polycarbosilanes must satisfy the conditions of $[4 < p + q + r + s < 100,000]$, and q and r can collectively or independently be zero.

The polycarbosilane may be produced from starting materials that are presently commercially available from many manufacturers and by using conventional polymerization processes. As an example of synthesis of the polycarbosilanes, the starting materials may be produced from common organo silane compounds or from polysilane as a starting material by heating an admixture of polysilane with polyborosiloxane in an inert atmosphere to thereby produce the corresponding polymer or by heating an admixture of polysilane with a low molecular weight carbosilane in an inert atmosphere to thereby produce the corresponding polymer or by heating an admixture of polysilane with a low molecular carbosilane in an inert atmosphere and in the presence of a catalyst such as polyborodiphenylsiloxane to thereby produce the corresponding polymer. Polycarbosilanes may also be synthesized by Grignard Reaction reported in U.S. Patent 5,153,295 hereby incorporated by reference.

An example of a preferred adhesion promoter having hydroxyl groups is phenol-formaldehyde resins or oligomers of the Formula XXVI: $[R_{18}C_6H_2(OH)(R_{19})]_u$ - where R_{18} is substituted or unsubstituted alkylene, cycloalkylene, vinyl, allyl, or aryl; R_{19} is alkyl, alkylene, vinylene, cycloalkylene,

allylene, or aryl; and $u = 3-100$. Examples of useful alkyl groups include $-C H_2-$ and $-(CH_2)_v-$ where $v > 1$. A particularly useful phenol-formaldehyde resin oligomer has a molecular weight of 1500 and is commercially available from Schenectady International Inc.

5

The present adhesion promoter is preferably added in small, effective amounts from about 0.5% to up to 20% based on the weight of the present thermosetting composition and amounts up to about 5.0 % by weight of the composition are generally more preferred.

10

By combining the adhesion promoter with the thermosetting component and subjecting the composition to thermal or a high energy source, the resulting compositions have superior adhesion characteristics throughout the entire polymer so as to ensure affinity to any contacted surface of the coating. The present adhesion promoters also improve striation control, viscosity, and film uniformity. Visual inspection confirms the presence of improved striation control.

The present compositions may also comprise additional components such as additional adhesion promoters, antifoam agents, detergents, flame retardants, pigments, plasticizers, stabilizers, and surfactants.

20

Porogen:

The term "pore" as used herein includes void and cells in a material and any other term meaning space occupied by gas in the material. Appropriate gases include relatively pure gases and mixtures thereof. Air, which is predominantly a mixture of N_2 and O_2 , is commonly distributed in the pores but pure gases such as nitrogen, helium, argon, CO_2 , or CO are also contemplated. Pores are typically spherical but may alternatively or additionally include tubular, lamellar, discoidal, voids having other shapes, or a combination of the preceding

25

shapes and may be open or closed. The term "porogen" as used herein means a decomposable material that is radiation, thermally, chemically, or moisture decomposable, degradable, depolymerizable, or otherwise capable of breaking down and includes solid, liquid, or gaseous material. The decomposed porogen is removable from or can volatilize or diffuse through a partially or fully cross-linked matrix to create pores in a subsequently fully cured matrix and thus, lower the matrix's dielectric constant and includes sacrificial polymers. Supercritical materials such as CO₂ may be used to remove porogen and decomposed porogen fragments. Preferably, for a thermally decomposable porogen, the porogen comprises a material having a decomposition temperature less than the glass transition temperature (T_g) of a material combined with it and greater than the curing temperature of the material combined with it. Preferably, the present porogens have a degradation or decomposition temperature of about 350°C or greater. Preferably, the degraded or decomposed porogens volatilize at a temperature greater than the cure temperature of the material with which the porogen is combined and less than the T_g of the material. Preferably, the degraded or decomposed porogens volatilize at a temperature of about 96°C or greater.

The phrase "porogen bonds to the thermosetting component" covers addition reactions, nucleophilic and electrophilic substitutions or eliminations, radical reactions, etc. Further alternative reactions may also include the formation of non-covalent bonds, such as Van der Waals, electrostatic bonds, ionic bonds, and hydrogen bonds.

Although International Patent Publication WO 00/31183 teaches that a porogen may be added to thermosettable benzocyclobutene, polyarylene, or thermosettable perfluoroethylene monomer to increase porosity thereof and thus, lower the dielectric constant of that resin, the reference teaches that a porogen that is known to function well with a first matrix system will not necessarily function well with another matrix system.

The present porogens preferably comprise unsubstituted polynorbornene, substituted polynorbornene, polycaprolactone, unsubstituted polystyrene, substituted polystyrene, polyacenaphthylene homopolymer, and polyacenaphthylene copolymer. The more preferred porogen is substituted
5 polynorbornene. Preferably, the porogen has functional groups selected from the group consisting of epoxy, hydroxy, carboxylic acid groups, amino, and ethynyl. Preferably, the porogen has a functional group on at least one of its ends.

Preferably, the porogen is bonded to the thermosetting component
10 through an ethynyl containing group. In one embodiment, the ethynyl containing group is first reacted with the porogen as shown in Figures 5, 7, 9, and 11. In a preferred embodiment, the ethynyl containing group is first reacted with the thermosetting component as shown in Figures 4, 6, 8, and 10. In Figures 4
15 though 11, although only 1,3/4-bis[1',3',5'-tris(3"/4"-bromophenyl)adamant-7'-yl]benzene is shown, it is understood that similar reaction occur for other useful thermosetting components including 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane and 1,3-bis{3'/4'-[1",3",5"-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis{3''''/4''''-
20 bromophenyl}adamantane. Useful ethynyl containing groups include fluorine; amine; or hydroxy; and preferably, are acetylene; 4-ethynylaniline; 3-hydroxyphenylacetylene; 4-fluorophenylacetylene; and 1-ethylcyclohexylamine. Preferably, a covalent bond forms between the porogen and the thermosetting component through the ethynyl containing group.

Useful polyacenaphthylene homopolymers may have weight average
25 molecular weights ranging from preferably about 300 to about 20,000; more preferably about 300 to about 10,000; and most preferably about 300 to about 7,000.

The amount of thermosetting component used is about 50 to about 90 weight percent while the amount of porogen used is about 10 to about 50
30 weight percent. Preferably, an adhesion promoter as described above is added

to the porogen bonded to the thermosetting component. Based on a composition comprising the adhesion promoter and the porogen bonded to the thermosetting component, about 0.1 to about 15 weight percent of adhesion promoter is used and about 5 to about 50 weight percent porogen bonded to the thermosetting component is used.

Pore Generation:

The term "degrade" as used herein refers to the breaking of covalent bonds. Such breaking of bonds may occur in numerous ways including heterolytic and homolytic breakage. The breaking of bonds need not be complete, i.e., not all breakable bonds must be cleaved. Furthermore, the breaking of bonds may occur in some bonds faster than in others. Ester bonds, for example, are generally less stable than amide bonds, and therefore, are cleaved at a faster rate. Breakage of bonds may also result in the release of fragments differing from one another, depending on the chemical composition of the degraded portion.

In the pore generation process, for thermally degradable porogens, thermal energy is applied to the porogen bonded to the thermosetting component to substantially degrade or decompose the porogen into its starting components or monomers. As used herein, "substantially degrade" preferably means at least 80 weight percent of the porogen degrades or decomposes. For the preferred thermosetting components of Formulae I and II above, the T_g is from about 400°C to about 450°C so the present porogens which have a degradation or decomposition temperature of about 350°C or greater are particularly useful with this thermosetting component.

Thermal energy is also applied to volatilize the substantially degraded or decomposed porogen out of the thermosetting component matrix. Preferably, the same thermal energy is used for both the degradation and volatilization

steps. As the amount of volatilized degraded porogen increases, the resulting porosity of the thermosetting component increases. For the preferred thermosetting components of Formulae I and II above, the T_g is from about 400°C to about 450°C so the present substantially degraded porogens which
5 have a volatilization temperature of about 96°C or greater are particularly useful with said thermosetting component.

Preferably, the cure temperature used for cross-linking the thermosetting component will also substantially degrade the porogen and volatilize it out of the
10 thermosetting matrix. Typical cure temperature and conditions will be described in the Utility section below.

The resulting pores may be uniformly or randomly dispersed throughout the matrix. Preferably, the pores are uniformly dispersed throughout the matrix.

Alternatively, other procedures or conditions which at least partially
15 remove the porogen without adversely affecting the thermosetting component may be used. Preferably, the porogen is substantially removed. Typical removal methods include, but are not limited to, exposure to radiation, such as but not limited to, electromagnetic radiation such as ultraviolet, x-ray, laser, or infrared radiation; mechanical energy such as sonication or physical pressure; or particle
20 radiation such as gamma ray, alpha particles, neutron beam, or electron beam.

Utility:

The term "layer" as used herein includes film and coating.

The term "low dielectric constant polymer" as used herein refers to an
25 organic, organometallic, or inorganic polymer with a dielectric constant of approximately 3.0, or lower. The low dielectric material is typically manufactured in the form of a thin layer having a thickness from 100 to 25,000 Angstroms but also may be used as thick films, blocks, cylinders, spheres etc.

The present composition of thermosetting component, adhesion promoter, and porogen is useful in lowering the dielectric constant of a material. Preferably, the dielectric material has a dielectric constant k of less than or equal to about 3.0 and more preferably, from about 1.9 to 3.0. The dielectric material
5 has a glass transition temperature of preferably at least about 350°C.

Layers of the instant compositions of thermosetting component, adhesion promoter, and porogen may be formed by solution techniques such as spraying, rolling, dipping, spin coating, flow coating, or casting, with spin coating being
10 preferred for microelectronics. Preferably, the present composition is dissolved in a solvent. Suitable solvents for use in such solutions of the present compositions include any suitable pure or mixture of organic, organometallic, or inorganic molecules that are volatilized at a desired temperature. Suitable
15 solvents include aprotic solvents, for example, cyclic ketones such as cyclopentanone, cyclohexanone, cycloheptanone, and cyclooctanone; cyclic amides such as N-alkylpyrrolidinone wherein the alkyl has from about 1 to 4 carbon atoms; and N-cyclohexylpyrrolidinone and mixtures thereof. A wide variety of other organic solvents may be used herein insofar as they are able to aid dissolution of the adhesion promoter and at the same time effectively control
20 the viscosity of the resulting solution as a coating solution. Various facilitating measures such as stirring and/or heating may be used to aid in the dissolution. Other suitable solvents include methylethylketone, methylisobutylketone, dibutyl ether, cyclic dimethylpolysiloxanes, butyrolactone, γ -butyrolactone, 2-heptanone, ethyl 3-ethoxypropionate, polyethylene glycol [di]methyl ether,
25 propylene glycol methyl ether acetate (PGMEA), and anisole, and hydrocarbon solvents such as mesitylene, xylenes, benzene, and toluene. Preferred solvent is cyclohexanone. Typically, layer thicknesses are between 0.1 to about 15 microns. As a dielectric interlayer for microelectronics, the layer thickness is generally less than 2 microns.

30

The present composition may be used in electrical devices and more specifically, as an interlayer dielectric in an interconnect associated with a single integrated circuit ("IC") chip. An integrated circuit chip typically has on its surface a plurality of layers of the present composition and multiple layers of metal conductors. It may also include regions of the present composition between discrete metal conductors or regions of conductor in the same layer or level of an integrated circuit.

In application of the instant polymers to ICs, a solution of the present composition is applied to a semiconductor wafer using conventional wet coating processes such as, for example, spin coating; other well known coating techniques such as spray coating, flow coating, or dip coating may be employed in specific cases. As an illustration, a cyclohexanone solution of the present composition is spin-coated onto a substrate having electrically conductive components fabricated therein and the coated substrate is then subjected to thermal processing. An exemplary formulation of the instant composition is prepared by dissolving the present composition in cyclohexanone solvent under ambient conditions with strict adherence to a clean-handling protocol to prevent trace metal contamination in any conventional apparatus having a non-metallic lining. The resulting solution comprises based on the total solution weight, from preferably about 2 to about 30 weight percent of thermosetting component, adhesion promoter, and porogen, and about 70 to about 98 weight percent solvent and more preferably about 5 to about 25 weight percent of thermosetting component, adhesion promoter, and porogen, and about 75 to about 95 weight percent solvent.

An illustration of the use of the present invention follows. Application of the instant compositions to form a layer onto planar or topographical surfaces or substrates may be carried out by using any conventional apparatus, preferably a spin coater, because the compositions used herein have a controlled viscosity suitable for such a coater. Evaporation of the solvent by any suitable means, such as simple air drying during spin coating, by exposure to an ambient environment, or by heating on a hot plate up to 350°C, may be employed. The

substrate may have on it at least one layer of the present preferred composition of thermosetting component, adhesion promoter, and porogen.

Substrates contemplated herein may comprise any desirable substantially solid material. Particularly desirable substrate layers comprise films, glass, ceramic, plastic, metal or coated metal, or composite material. In preferred embodiments, the substrate comprises a silicon or gallium arsenide die or wafer surface, a packaging surface such as found in a copper, silver, nickel or gold plated leadframe, a copper surface such as found in a circuit board or package interconnect trace, a via-wall or stiffener interface ("copper" includes considerations of bare copper and its oxides), a polymer-based packaging or board interface such as found in a polyimide-based flex package, lead or other metal alloy solder ball surface, glass and polymers. Useful substrates include silicon, silicon nitride, silicon oxide, silicon oxycarbide, silicon dioxide, silicon carbide, silicon oxynitride, titanium nitride, tantalum nitride, tungsten nitride, aluminum, copper, tantalum, organosiloxanes, organo silicon glass, and fluorinated silicon glass. In other embodiments, the substrate comprises a material common in the packaging and circuit board industries such as silicon, copper, glass, and polymers. The present compositions may also be used as a dielectric substrate material in microchips, multichip modules, laminated circuit boards, or printed wiring boards. The circuit board made up of the present composition will have mounted on its surface patterns for various electrical conductor circuits. The circuit board may include various reinforcements, such as woven non-conducting fibers or glass cloth. Such circuit boards may be single sided, as well as double sided.

Layers made from the present compositions possess a low dielectric constant, high thermal stability, high mechanical strength, and excellent adhesion to electronic substrate surfaces. Because the adhesion promoter is molecularly dispersed, these layers demonstrate excellent adhesion to all affixed surfaces including underlying substrates and overlaid capping or masking layers,

such as SiO_2 and Si_3N_4 capping layers. The use of these layers eliminates the need for an additional process step in the form of at least one primer coating application to achieve adhesion of the film to a substrate and/or overlaid surface.

5

After application of the present composition to an electronic topographical substrate, the coated structure is subjected to a bake and cure thermal process at increasing temperatures ranging from about 50°C up to about 450°C to polymerize the coating. The curing temperature is at least about 300°C because
10 a lower temperature is insufficient to complete the reaction herein. Generally, it is preferred that curing is carried out at temperatures of from about 375°C to about 425°C . Curing may be carried out in a conventional curing chamber such as an electric furnace, hot plate, and the like and is generally performed in an inert (non-oxidizing) atmosphere (nitrogen) in the curing chamber. In addition to
15 furnace or hot plate curing, the present compositions may also be cured by exposure to ultraviolet radiation, microwave radiation, or electron beam radiation as taught by commonly assigned patent publication PCT/US96/08678 and US Patents 6,042,994; 6,080,526; 6,177,143; and 6,235,353, which are incorporated herein by reference in their entireties. Any non oxidizing or
20 reducing atmospheres (e.g., argon, helium, hydrogen, and nitrogen processing gases) may be used in the practice of the present invention, if they are effective to conduct curing of the present adhesion promoter-modified thermosetting component to achieve the low k dielectric layer herein.

25 While not to be construed as limiting, it is observed that the processing used to prepare the present low dielectric constant composition results in a homogeneous solution of thermosetting component, adhesion promoter, and porogen. The preferred silane adhesion promoter advantageously serves multiple functions in the low dielectric constant composition. For example, the
30 processing of the present composition enables the preferred polycarbosilane adhesion promoter to interact with both the porogen and the unsaturated

structures of thermosetting component. It is believed that the silane portions of the preferred polycarbosilane interact with the porogen and thermosetting component. It is speculated that the polycarbosilane acts as a surfactant or emulsification agent to uniformly disperse the porogen within the thermosetting component in the low dielectric composition. This is critical to producing a composition that gives a homogeneous film (or layer) with uniformly dispersed pores of very small dimension. The silane portion of the polycarbosilane also reacts with the substrate surfaces, thereby creating a chemically bonded adherent interface for the dominant thermosetting monomer precursor. It has been proposed that silylene/silyl radicals being available throughout the composition act as attachment sources to fasten and secure any interface surface of contact by chemical bonding therewith. The interactions between the various components and the reactions of the silane portion may occur during formulation and treatment prior to layer formation. As indicated, the dispersion of silane functionality with the porogen and thermosetting component throughout the composition accounts for the uniform porosity in the resulting layers. The dispersion of the silane functionality also leads to reactive radicals as well as the superb adhesion of the instant layers to both underlying substrate surfaces as well as overlayered surface structures such as cap or masking layers. Crucial to the materials discovered herein are the findings that the preferred Formula XXV polycarbosilane adhesion promoters have a hydrido substituted silicon in the backbone structure of the polycarbosilane. This feature of the polycarbosilane enables it to: (1) mix uniformly with the porogen to form a homogeneous composition, (2) be reactive with thermosetting component; (3) uniformly blend and disperse the porogen within the thermosetting component providing a uniform composition leading to uniform distribution of small pores in the final porous layer, and (4) generate a polycarbosilane-modified thermosetting composition and porous layer that possesses improved adhesion performance.

As indicated earlier, the present adhesion promoter-modified thermosetting component (a) coating may act as an interlayer and be covered by other coatings, such as other dielectric (SiO_2) coatings, SiO_2 modified ceramic oxide layers, silicon containing coatings, silicon carbon containing coatings, silicon nitrogen containing coatings, silicon-nitrogen-carbon containing coatings, diamond like carbon coatings, titanium nitride coatings, tantalum nitride coatings, tungsten nitride coatings, aluminum coatings, copper coatings, tantalum coatings, organosiloxane coatings, organo silicon glass coatings, and fluorinated silicon glass coatings. Such multilayer coatings are taught in U.S. Pat. No. 4,973,526, which is incorporated herein by reference. And, as amply demonstrated, the present polycarbosilane-modified thermosetting component (a) prepared in the instant process may be readily formed as interlined dielectric layers between adjacent conductor paths on fabricated electronic or semiconductor substrates.

The present films may be used in dual damascene (such as copper) processing and subtractive metal (such as aluminum or aluminum/tungsten) processing for integrated circuit manufacturing. The present compositions may be used as an etch stop, hardmask, air bridge, or passive coating for enveloping a completed wafer. The present composition may be used in a desirable all spin-on stacked film as taught by Michael E. Thomas, "Spin-On Stacked Films for Low k_{eff} Dielectrics", *Solid State Technology* (July 2001), incorporated herein in its entirety by reference. The present layers may be used in stacks with other layers comprising organosiloxanes such as taught by commonly assigned US Patent 6,143,855 and pending US Serial No. 10/078919 filed February 19, 2002; Honeywell International Inc.'s commercially available HOSP[®] product; nanoporous silica such as taught by commonly assigned US Patent 6,372,666; Honeywell International Inc.'s commercially available NANOGLASS[®] E product; organosilsesquioxanes taught by commonly assigned WO 01/29052; and fluorosilsesquioxanes taught by commonly assigned WO 01/29141, incorporated herein in their entirety.

Analytical Test Methods:

Proton NMR: A 2-5 mg sample of the material to be analyzed was put into an NMR tube. About 0.7 ml deuterated chloroform was added. The mixture was shaken by hand to dissolve the material. The sample was then analyzed using a Varian 400MHz NMR.

High Performance Liquid Chromatography (HPLC): A HPLC with a Phenomenex luna Phenyl-Hexyl 250X4.6mm 5 micron column was used. The column temperature was set at 40°C. Water and acetonitrile were used to improve peak separation.

TIME	WATER	ACETONITRILE
Initial	20%	80%
10 minutes	0%	100%
30 minutes	0%	100%

The following experimental conditions were used:

INJECTION VOLUME	10 microliters
DETECTION	UV at 200nm
STOP TIME	30 minutes
POST TIME	5 minutes

The samples were prepared as follows.

For a mixture of the halogenated intermediate such as the mixture of 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane and 1,3/4-bis[1',3',5'-tris(3''/4''-bromophenyl)adamant-7'-yl]benzene of Preparation 1 below, the reaction mixture (0.5-1 milliliter) was shaken with approximately 4% HCl (several milliliters). The organic layer was shaken with water. An organic layer sample (twenty microliters) was taken and added to acetonitrile (one milliliter).

For a mixture of the final product such as the mixture of 1,3,5,7-tetrakis[3'/4'-(phenylethynyl)phenyl]adamantane and 1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-yl}benzene of Preparation 1 below, the reaction mixture (0.5 gram) was mixed with chloroform (five milliliters) and

3-5% HCl (5 milliliters) and shaken. The organic layer was washed by water. An organic layer sample (100 microliters) was added to tetrahydrofuran (0.9 milliliter).

- 5 Gel Permeation Chromatography (GPC 1): The GPC analysis was done with Waters liquid chromatography system composed from Water 717 plus Autosampler, Waters in-line degasser, Waters 515 HPLC pump, Waters 410 Differential Refractometer (RI detector), and two columns: HP PI gel 5 μ MIXED D. The analysis conditions were:

10

Mobile Phase	Tetrahydrofuran (THF)
Column flow (milliliters/min)	1.0
Column temperature (°C)	40.0
Detection	Refractive Index, Polarity positive
Analysis run time	25 min
Injection quantity (μ L)	50

The solid product sample (10 milligrams) was prepared by dissolving in tetrahydrofuran (one milliliter).

- 15 Gel Permeation Chromatography (GPC 2): This method may be used to provide additional detail about the dimer and trimer peaks. The following conditions are used:

Analysis Apparatus	Shimadzu LC10
Separation Column	Plgel 5 μ pre-column
	Plgel 5 μ 1000Å, 300x7.5nm
	Plgel 5 μ 500Å, 300x7.5nm
	Plgel 5 μ 100Å, 300x7.5nm
Mobile Phase	Eluent A: toluene
Column Flow (milliliters/minute)	1.0
Column Temperature (°C)	40
Detection	Refractive Index, Polarity Positive
Analysis Run Time (minutes)	32
Trial Solution	10 milligrams/milliliter toluene
Injection Quantity (μ)	50

To calculate the contents, in other words area-%, the peak area belonging to the monomer or the peak area belonging to the oligomer or polymer is related to the total of all the peak areas in the chromatogram.

5 Gel Permeation Chromatography (GPC 3): Separation was performed with a Waters 2690 separation module with Waters 996 diode array and Waters 410 differential refractometer detectors. The separation was performed on two PLgel 3 μ m Mixed-E 300 x 7.5 mm columns with chloroform flowing at 1 ml/min. Injection volumes of 25 μ l of solutions of about 1 mg/ml concentration were run
10 in duplicate. Good reproducibility was observed.

The column was calibrated with relatively monodisperse polystyrene standards between 20,000 and 500 molecular weight. With the lower molecular weight standards nine distinct components could be resolved
15 corresponding to butyl terminated styrene monomer through oligomers with nine styrenes. The logs of the peak molecular weight of the standards were fit with a third order polynomial of the elution time. The instrumental broadening was evaluated from the ratio of the full width at half maximum to the mean elution time of toluene.

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The absorbance for Preparations 1 and 2 below was a maximum at about 284nm. The chromatograms had similar shapes at absorbance at wavelengths below about 300 nm. The results presented here correspond to 254 nm absorbance. The peaks were identified by the molecular weight of the
25 polystyrene that would be eluting at the same time. These values should not be considered as measurements of molecular weight of the Preparation 1 and 2 oligomers. The sequential elution of higher oligomers, trimers, dimers, oligomers, and incomplete oligomers at increasing times can be quantitated.

30 Each component was broader than that which would be observed for a monodisperse species. This width was analyzed from the full width in minutes

at half maximum of the peak. To roughly account for the instrumental broadening, we calculated

$$\text{width}_{\text{corrected}} = [\text{width}_{\text{observed}}^2 - \text{width}_{\text{instrument}}^2]^{1/2}$$

5

where $\text{width}_{\text{instrument}}$ is the observed width of toluene corrected by the ratio of the elution times of the peak to that for toluene. The peak width was converted to a molecular weight width through the calibration curve and ratioed to the peak molecular width. Since the molecular weight of styrene oligomers was proportional to the square of their size, the relative molecular weight width can be converted to a relative oligomer size width by dividing by 2. This procedure accounted for the difference in molecular configuration of the two species.

10

15 Liquid Chromatography-Mass Spectroscopy (LC-MS): This analysis was performed on a Finnigan/MAT TSQ7000 triple stage quadrupole mass spectrometer system, with an Atmospheric Pressure Ionization (API) interface unit, using a Hewlett-Packard Series 1050 HPLC system as the chromatographic inlet. Both mass spectral ion current and variable single wavelength UV data were acquired for time-intensity chromatograms.

20

Chromatography was conducted on a Phenomenex Luna 5-micron phenylhexyl column (250x4.6mm). Sample auto-injections were generally between 5 and 20 microliters of concentrated solutions, both in tetrahydrofuran and without tetrahydrofuran. The preferred preparation of concentrated sample solutions for analysis was dissolution in tetrahydrofuran, of about 5 milligrams solid product per milliliter, for 10 microliter injections. The mobile phase flow through the column was 1.0 milliliter/minute of acetonitrile/water, initially 70/30 for 1 minute then gradient programmed to 100% acetonitrile at 10 minutes and held until 40 minutes.

25

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Atmospheric Pressure Chemical Ionization (APCI) mass spectra were

recorded in both positive and negative ionization, in separate experiments. Positive APCI was more informative of molecular structure for these final products, providing protonated pseudomolecular ions including adducts with acetonitrile matrix. The APCI corona discharge was 5 microamps, about 5kV for positive ionization, and about 4kV for negative ionization. The heated capillary line was maintained at 200°C and the vaporizer cell at 400°C. The ion detection system after quadrupole mass analysis was set at 15kV conversion dynode and 1500V electron multiplier voltage. Mass spectra were typically recorded at 1.0 second/scan from about m/z 50 to 2000 a.m.u. for negative ionization, and from about m/z 150 a.m.u. up for positive ionization. In separate positive ion experiments, the mass range was scanned up both to 2000 a.m.u. in low mass tune/calibration mode and to 4000 a.m.u. in high mass tune/calibration mode.

Differential Scanning Calorimetry (DSC): DSC measurements were performed using a TA Instruments 2920 Differential Scanning Calorimeter in conjunction with a controller and associated software. A standard DSC cell with temperature ranges from 250°C to 725°C (inert atmosphere: 50 ml/min of nitrogen) was used for the analysis. Liquid nitrogen was used as a cooling gas source. A small amount of sample (10-12 mg) was carefully weighed into an Auto DSC aluminum sample pan (Part # 990999-901) using a Mettler Toledo Analytical balance with an accuracy of ± 0.0001 grams. Sample was encapsulated by covering the pan with the lid that was previously punctured in the center to allow for outgasing. Sample was heated under nitrogen from 0°C to 450°C at a rate of 100°C/minute (cycle 1), then cooled to 0°C at a rate of 100°C/minute. A second cycle was run immediately from 0°C to 450°C at a rate of 100°C/minute (repeat of cycle 1). The cross-linking temperature was determined from the first cycle.

FTIR analysis: FTIR spectra were taken using a Nicolet Magna 550 FTIR spectrometer in transmission mode. Substrate background spectra were taken on uncoated substrates. Film spectra were taken using the substrate as

background. Film spectra were then analyzed for change in peak location and intensity.

Dielectric Constant: The dielectric constant was determined by coating a thin film of aluminum on the cured layer and then doing a capacitance-voltage measurement at 1MHz and calculating the k value based on the layer thickness.

Glass Transition Temperature (T_g): The glass transition temperature of a thin film was determined by measuring the thin film stress as a function of temperature. The thin film stress measurement was performed on a KLA 3220 Flexus. Before the film measurement, the uncoated wafer was annealed at 500°C for 60 minutes to avoid any errors due to stress relaxation in the wafer itself. The wafer was then deposited with the material to be tested and processed through all required process steps. The wafer was then placed in the stress gauge, which measured the wafer bow as function of temperature. The instrument calculated the stress versus temperature graph, provided that the wafer thickness and the film thickness were known. The result was displayed in graphic form. To determine the T_g value, a horizontal tangent line was drawn (a slope value of zero on the stress vs. temperature graph). T_g value was where the graph and the horizontal tangent line intersect.

It should be reported if the T_g was determined after the first temperature cycle or a subsequent cycle where the maximum temperature was used because the measurement process itself may influence T_g.

Isothermal Gravimetric Analysis (ITGA) Weight Loss: Total weight loss was determined on the TA Instruments 2950 Thermogravimetric Analyzer (TGA) used in conjunction with a TA Instruments thermal analysis controller and associated software. A Platinel II Thermocouple and a Standard Furnace with a temperature range of 25°C to 1000°C and heating rate of 0.1°C to 100°C/min were used. A small amount of sample (7 to 12 mg) was weighed on the TGA's balance (resolution: 0.1 g; accuracy: to ±0.1%) and heated on a platinum pan.

Samples were heated under nitrogen with a purge rate of 100 ml/min (60 ml/min going to the furnace and 40 ml/min to the balance). Sample was equilibrated under nitrogen at 20°C for 20 minutes, then temperature was raised to 200°C at a rate of 10°C/minute and held at 200°C for 10 minutes. Temperature was then
5 ramped to 425°C at a rate of 10°C/minute and held at 425°C for 4 hours. The weight loss at 425°C for the 4 hour period was calculated.

Shrinkage: Film shrinkage was measured by determining the film thickness before and after the process. Shrinkage was expressed in percent of
10 the original film thickness. Shrinkage was positive if the film thickness decreased. The actual thickness measurements were performed optically using a J.A. Woollam M-88 spectroscopic ellipsometer. A Cauchy model was used to calculate the best fit for Psi and Delta (details on Ellipsometry can be found in e.g. "Spectroscopic Ellipsometry and Reflectometry" by H.G. Thompson and
15 William A. McGahan, John Wiley and Sons, Inc., 1999).

Refractive Index: The refractive index measurements were performed together with the thickness measurements using a J.A. Woollam M-88 spectroscopic ellipsometer. A Cauchy model was used to calculate the best fit
20 for Psi and Delta. Unless noted otherwise, the refractive index was reported at a wavelength of 633nm (details on Ellipsometry can be found in e.g. "Spectroscopic Ellipsometry and Reflectometry" by H.G. Thompson and William A. McGahan, John Wiley and Sons, Inc., 1999).

Modulus and Hardness: Modulus and hardness were measured using
25 instrumented indentation testing. The measurements were performed using a MTS Nanoindenter XP (MTS Systems Corp., Oak Ridge, TN). Specifically, the continuous stiffness measurement method was used, which enabled the accurate and continuous determination of modulus and hardness rather than
30 measurement of a discrete value from the unloading curves. The system was calibrated using fused silica with a nominal modulus of 72 +/- 3.5 GPa. The modulus for fused silica was obtained from average value between 500 to 1000

nm indentation depth. For the thin films, the modulus and hardness values were obtained from the minimum of the modulus versus depth curve, which is typically between 5 to 15% of the film thickness.

5 Tape Test: The tape test was performed following the guidelines given in ASTM D3359-95. A grid was scribed into the dielectric layer according to the following. A tape test was performed across the grid marking in the following manner: (1) a piece of adhesive tape, preferably Scotch brand #3m600-1/2X1296, was placed on the present layer, and pressed down firmly to make
10 good contact; and (2) the tape was then pulled off rapidly and evenly at an angle of 180° to the layer surface. The sample was considered to pass if the layer remained intact on the wafer, or to have failed if part or all of the film pulled up with the tape.

15 Stud Pull Test: Epoxy-coated studs were attached to the surface of a wafer containing the layers of the present invention. A ceramic backing plate was applied to the back side of the wafer to prevent substrate bending and undue stress concentration at the edges of the stud. The studs were then pulled in a direction normal to the wafer surface by a testing apparatus employing
20 standard pull protocol steps. The stress applied at the point of failure and the interface location were then recorded.

Compatibility with Solvents: Compatibility with solvents was determined by measuring film thickness, refractive index, FTIR spectra, and dielectric
25 constant before and after solvent treatment. For a compatible solvent, no significant change should be observed.

Average Pore Size Diameter: The N₂ isotherms of porous samples was measured on a Micromeritics ASAP 2000 automatic isothermal N₂ sorption
30 instrument using UHP (ultra high purity industrial gas) N₂, with the sample immersed in a sample tube in a liquid N₂ bath at 77°K.

For sample preparation, the material was first deposited on silicon wafers using standard processing conditions. For each sample, three wafers were prepared with a film thickness of approximately 6000 Angstroms. The films were then removed from the wafers by scraping with a razor blade to generate powder samples. These powder samples were pre-dried at 180°C in an oven before weighing them, carefully pouring the powder into a 10 mm inner diameter sample tube, then degassing at 180 °C at 0.01 Torr for > 3 hours.

The adsorption and desorption N₂ sorption was then measured automatically using a 5 second equilibration interval, unless analysis showed that a longer time was required. The time required to measure the isotherm was proportional to the mass of the sample, the pore volume of the sample, the number of data points measured, the equilibration interval, and the P/Po tolerance. (P is actual pressure of the sample in the sample tube. Po is the ambient pressure outside the instrument.) The instrument measures the N₂ isotherm and plots N₂ versus P/Po.

The apparent BET (Brunauer, Emmett, Teller method for multi-layer gas absorption on a solid surface disclosed in S. Brunauer, P. H. Emmett, E. Teller; *J. Am. Chem. Soc.*, 60, 309-319 (1938)) surface area was calculated from the lower P/Po region of the N₂ adsorption isotherm using the BET theory, using the linear section of the BET equation that gives an R² fit > 0.9999.

The pore volume was calculated from the volume of N₂ adsorbed at the relative pressure P/Po value, usually P/Po ~ 0.95, which is in the flat region of the isotherm where condensation is complete, assuming that the density of the adsorbed N₂ is the same as liquid N₂ and that all the pores are filled with condensed N₂ at this P/Po.

The pore size distribution was calculated from the adsorption arm of the N₂ isotherm using the BJH (E. P. Barret, L. G. Joyner, P. P. Halenda; *J. Am.*

Chem. Soc., 73, 373-380 (1951)) pore size distribution from the N₂ isotherm using the Kelvin equation) theory. This uses the Kelvin equation, which relates curvature to suppression of vapor pressure, and the Halsey equation, which describes the thickness of the adsorbed N₂ monolayer versus P/P₀, to convert
5 the volume of condensed N₂ versus P/P₀ to the pore volume in a particular range of pore sizes.

The average cylindrical pore diameter D was the diameter of a cylinder that has the same apparent BET surface area S_a (m²/g) and pore volume V_p
10 (cc/g) as the sample, so $D \text{ (nm)} = 4000V_p/S_a$.

Thermal Desorption Mass Spectroscopy: Thermal Desorption Mass Spectroscopy (TDMS) is used to measure the thermal stability of a material by
15 analyzing the desorbing species while the material is subjected to a thermal treatment.

The TDMS measurement was performed in a high vacuum system equipped with a wafer heater and a mass spectrometer, which was located
20 close to the front surface of the wafer. The wafer was heated using heating lamps, which heat the wafer from the backside. The wafer temperature was measured by a thermocouple, which was in contact with the front surface of the wafer. Heater lamps and thermocouple were connected to a programmable temperature controller, which allowed several temperature ramp and soak
25 cycles. The mass spectrometer was a Hiden Analytical HAL IV RC RGA 301. Both mass spectrometer and the temperature controller were connected to a computer, which read and recorded the mass spectrometer and the temperature signal versus time.

30 To perform TDMS analysis, the material was first deposited as a thin film onto an 8 inch wafer using standard processing methods. The wafer was then placed in the TDMS vacuum system and the system was pumped down to a

pressure below 1e-7 torr. The temperature ramp was then starting using the temperature controller. The temperature and the mass spectrometer signal were recorded using the computer. For a typical measurement with a ramp rate of about 10 degree C per minute, one complete mass scan and one temperature measurement are recorded every 20 seconds. The mass spectrum at a given time and temperature at a given time can be analyzed after the measurement is completed.

Examples:

Comparative A:

We measured the dielectric constant of a composition similar to Example 5 of our International Patent Publication WO 01/78110 and the dielectric constant was 2.7.

Preparations:

15

Preparation 1 - Preparation of Thermosetting Component (referred to herein as "P1")

Step (a): Preparation of Mixture of

1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane (shown in Figure 1A);
1,3/4-bis[1',3',5'-tris(3''/4''-bromophenyl)adamant-7'-yl]benzene (shown in Figure 1C); and at least 1,3-bis{3'/4'-[1'',3'',5''-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis(3''''/4''''-bromophenyl)adamantane (shown in Figure 1C) (collectively "P1 Step (a) Product")

25

A first reaction vessel was loaded with adamantane (200 grams), bromobenzene (1550 milliliters), and aluminum trichloride (50 grams). The reaction mixture was heated to 40°C by a thermostatted water bath. Tert-butyl bromide (1206 grams) was added slowly over a period of 4-6 hours to the reaction mixture. The reaction mixture at 40°C was stirred overnight.

30

A second reaction vessel was loaded with 1000 milliliters of aqueous hydrogen chloride (5%w/w). The contents of the first reaction vessel were gradually discharged into the second reaction vessel while maintaining the reaction mixture at 25-35°C by an external ice bath. An organic phase (dark brown lower phase) was separated and washed with water (1000 milliliters). About 1700 milliliters of the organic phase remained.

A third reaction vessel was loaded with 20.4 liters of petroleum ether (mainly isooctane with a boiling range of 80°C-110°C). The contents of the second reaction vessel were slowly added over a period of one hour to the third reaction vessel. The resulting mixture was stirred for at least one hour. The precipitate was filtered off and the filter cake was washed twice with 300 milliliters per wash of the aforementioned petroleum ether. The washed filter cake was dried overnight at 45°C at 40mbar. The P1 Step (a) Product yield was 407 grams dry weight. This reaction is shown in Figures 1A through 1C as follows. Figure 1A shows the resulting monomer. Figure 1B shows the resulting generic dimer and higher products while Figure 1C shows the resulting specific dimer and trimer covered by the Figure 1B structure.

Analytical techniques including GPC, HPLC, and NMR were used to identify the product. GPC analysis showed: 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane (shown in Figure 1A) had a peak molecular weight of 430; 1,3/4-bis[1',3',5'-tris(3''/4''-bromophenyl)adamant-7'-yl]benzene (shown in Figure 1C) had a peak molecular weight of 820; 1,3-bis{3'/4'-[1'',3'',5''-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis(3''''/4''''-bromophenyl)adamantane (shown in Figure 1C) had a peak molecular weight of about 1150 (shoulder).

Step (b): Preparation of Mixture of
1,3,5,7-tetrakis[3',4'-(phenylethynyl)phenyl]adamantane (shown in Figure 1D);
1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-yl} benzene
(shown in Figure 1F); and at least 1,3-bis{3'/4'-[1'',3'',5''-tris[3'''/4'''-(phenylethynyl)phenyl]adamant-7''-yl]phenyl}-5,7-bis[3''''/4''''-

(phenylethynyl)phenyl]adamantane (shown in Figure 1F) (collectively "P1 Step (b) Product")

A first reactor under nitrogen was loaded with toluene (1500 milliliters), triethylamine (4000 milliliters), and the P1 Step (a) Product prepared above (1000 grams dry). The mixture was heated to 80°C and bis-(triphenylphosphine)palladium(II)dichloride (i.e., $[\text{Ph}_3\text{P}]_2\text{PdCl}_2$) (7.5 grams) and triphenylphosphine (i.e. $[\text{Ph}_3\text{P}]$) (15 grams) were added. After ten minutes, copper(I)iodide (7.5 grams) was added.

Over a period of three hours, a solution of phenylacetylene (750 grams) was added to the first reactor. The reaction mixture at 80°C was stirred for 12 hours to ensure that the reaction was complete. Toluene (4750 milliliters) was added. The solvent was then distilled off under reduced pressure and a maximum sump temperature and the reaction mixture was cooled down to about 50°C. The triethylammonium bromide (about 1600 milliliters) was filtered off. The filter cake was washed three times with 500 milliliters per wash of toluene. The organic phase was washed with 1750 milliliters of HCl (10 w/w%) and then washed with water (2000 milliliters).

To the washed organic phase, water (1000 milliliters), ethylene diamine tetraacetic acid (EDTA) (100 grams), and dimethylglyoxime (20 grams) were added. About 150 milliliters of NH_4OH (25 w/w%) were added to achieve a pH of 9. The reaction mixture was stirred for one hour. The organic phase was separated and washed with water (1000 milliliters). With a Dean-Stark trap, azeotropic drying occurred until water evolution ceased. Filtering agent dolomite (100 grams) (tradename Tonsil) was added. The mixture was heated to 100°C for 30 minutes. Dolomite was filtered off with a cloth filter having fine pores and the remainder was washed with toluene (200 milliliters). Silica (100 grams) was added. The reaction mixture was stirred for 30 minutes. The silica was filtered off with a cloth filter having fine pores and the remainder was washed with toluene (200 milliliters). Aqueous NH_3 (20 w/w%), in an amount of 2500

milliliters, and 12.5g of N-acetylcysteine were added. The phases were separated. The organic phase was washed with 1000 milliliters of HCl (10% w/w) and then washed two times with 1000 milliliters per wash of water. The toluene was distilled off under a reduced pressure of about 120mbar. The pot
5 temperature did not exceed about 70°C. A dark brown viscous oil (1500-1700 milliliters) remained. To the hot mass in the pot, iso-butyl acetate (2500 milliliters) was added and a dark brown solution formed (4250 milliliters).

A second reactor was loaded with 17000 milliliters of petroleum ether
10 (mainly isooctane with a boiling range of 80°C-110°C). The contents of the first reactor were added over a period of one hour to the second reactor and stirred overnight. The precipitate was filtered and washed four times with 500 milliliters per wash of the aforescribed petroleum ether. The product was dried under reduced pressure for four hours at 45°C and five hours at 80°C.
15 The P1 Step (B) Product yield was 850-900 grams. This reaction is shown in Figures 1D through 1F as follows. Figure 1D shows the resulting monomer. Figure 1E shows the resulting generic dimer and higher products while Figure 1F shows the resulting specific dimer and trimer covered by the Figure 1F structure.

20 Analytical techniques including GPC, HPLC, NMR, and FTIR were used to identify the product. GPC analysis showed: 1,3,5,7-tetrakis[3',4'-(phenylethynyl)phenyl]adamantane (shown in Figure 1D) had a peak molecular weight of about 900; 1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-yl} benzene (shown in Figure 1F) had a peak
25 molecular weight of about 1500; 1,3-bis{3'/4'-[1'',3'',5''-tris[3'''/4'''-(phenylethynyl)phenyl]adamant-7''-yl]phenyl}-5,7-bis[3''''/4''''-(phenylethynyl)phenyl]adamantane (shown in Figure 1F) had a peak molecular weight of about 2100 (shoulder).

30 The melting point was 164-167°C. From NMR, a multiplet occurred at 6,9-8 ppm 2,8+-0,2H (aromatic part) and 1,7-2,7 ppm 1H+-0,2H (cage portion). From GPC, the ratio of the monomeric and small molecules to oligomeric compounds was 50±5%. FTIR showed the following:

PEAKS IN CENTIMETERS ⁻¹ (PEAK INTENSITY)	STRUCTURE
3050 (weak)	Aromatic C-H
2930 (weak)	Aliphatic C-H on adamantane
2200 (very weak)	Acetylene
1600 (very strong)	Aromatic C = C
1500 (strong)	
1450 (medium)	
1350 (medium)	

LC-MS study showed the presence of peaks:

- of main monomeric product m/z 840,
- of its derivatives, m/z 640, 740, 940; 706, 606, 806; 762, 662, 862; 938, 838, 1038
- of dimeric products, m/z 1402, 1302, 1502; 1326, 1226, 1426.

The GPC 3 results follow.

Amount (Weight %)					Peak Molecular Weight (Relative to PS)				ΔR/R
Inc Mnmr	Mnmr	Dmr	Trmr	Hi Olgr	Inc Mnmr	Mnmr	Dmr	Trmr	Mnmr
w/mnmr	49.3	30.9	10.1	9.7		744	1304	1676	0.10

Preparation 2 - Preparation of Thermosetting Component (referred to herein as "P2")

Step (a): Preparation of Mixture of
 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane (shown in Figure 1A);
 1,3/4-bis[1',3',5'-tris(3''/4''-bromophenyl)adamant-7'-yl] benzene (shown in
 Figure 1C); and at least 1,3-bis{3'/4'-[1'',3'',5''-tris(3'''/4'''-
 bromophenyl)adamant-7''-yl]phenyl}-5,7-bis(3''''/4''''-
 bromophenyl)adamantane(shown in Figure 1C)(collectively "P2 Step (a)
 Product")

A first reaction vessel was loaded with 1,4-dibromobenzene (587.4 grams) and aluminum trichloride (27.7 grams). This reaction mixture was heated to 90°C by a thermostatted water bath and maintained at this temperature for one hour without stirring and for an additional one hour with stirring. The reaction mixture was cooled down to 50°C. Adamantane (113.1 grams) was added to the cooled reaction mixture. Over a period of four hours, t-butyl-bromobenzene (796.3 grams) was added to the reaction mixture. The reaction mixture was stirred for an additional 12 hours.

A second reaction vessel was loaded with HCl (566 milliliters, 10% aqueous w/w). The contents of the first reaction vessel at 50°C were discharged into the second reaction vessel while maintaining the mixture at 25-35°C by an external ice bath. The reaction mass was a light brown suspension. The organic phase was a dark brown lower phase and separated from the reaction mixture. The separated organic phase was washed with water (380 milliliters). After this washing, about 800 milliliters of organic phase remained.

A third reaction vessel was loaded with heptane (5600 milliliters). Slowly over a period of one hour, the contents of the second reaction vessel were added to the third reaction vessel. The suspension was stirred for at least four hours and the precipitate was filtered off. The filter cake was washed twice with 300 milliliters per wash of heptane. The P2 Step (a) Product yield was 526.9 grams (wet) and 470.1 grams (dry).

Analytical techniques including GPC, HPLC, and NMR were used to identify the product. GPC analysis showed: 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane (shown in Figure 1A) had a peak molecular weight of about 430; 1,3/4-bis[1',3',5'-tris(3''/4''-bromophenyl)adamant-7'-yl] benzene (shown in Figure 1C) had a peak molecular weight of about 820; 1,3-bis{3'/4'-[1'',3'',5''-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis(3''''/4''''-bromophenyl)adamantane (shown in Figure 1C) had a peak molecular weight of about 1150 (shoulder).

Step (b): Preparation of Mixture of
1,3,5,7-tetrakis[3',4'-(phenylethynyl)phenyl]adamantane (shown in Figure 1D);
1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-yl}benzene
5 (shown in Figure 1F); and at least 1,3-bis{3'/4'-[1'',3'',5''-tris[3'''/4'''-(phenylethynyl)phenyl]adamant-7'''-yl]phenyl}-5,7-bis[3''''/4''''-(phenylethynyl)phenyl]adamantane (shown in Figure 1F)(collectively "P2 Step (b) Product")

10 A first reaction pot under nitrogen was loaded with toluene (698 milliliters), triethylamine (1860 milliliters), and the P2 Step (a) Product prepared above (465 grams dry). The mixture was heated to 80°C. Palladium-triphenylphosphine complex (i.e. $[\text{Ph}(\text{PPh}_3)_2\text{Cl}_2]$ (4.2 grams) was added to the reaction mixture. After waiting ten minutes, triphenylphosphine (i.e., PPh_3) (8.4
15 grams) was added to the reaction mixture. After waiting another ten minutes, copper(I)-iodide (4.2 grams) was added to the reaction mixture.

Over a period of three hours, a solution of phenylacetylene (348.8 grams) was added to the reaction mixture. The reaction mixture at 80°C was stirred for
20 12 hours to ensure that the reaction was complete. Toluene (2209 milliliters) was added to the reaction mixture and then distilled off under reduced pressure and a maximum sump temperature. The reaction mixture was cooled down to about 50°C and the triethylammonium bromide was filtered off. The filter cake was washed twice with 250 milliliters per wash of toluene. The organic phase
25 was washed with HCl (10 w/w%)(500 milliliters) and water (500 milliliters).

To the organic phase, water (500 milliliters), EDTA (18.6 grams), and dimethylglyoxime (3.7 grams) were added. NH_4OH (25 w/w%)(about 93 milliliters) was added to keep the pH = 9. The reaction mixture was stirred for
30 one hour. The organic phase was separated from the insoluble material and the emulsion containing the palladium-complex. The separated organic phase was washed with water (500 milliliters). With a Dean-Stark trap, azeotropic drying of the washed organic phase occurred until water evolution ceased. Filtering

agent dolomite (tradename Tonsil)(50 grams) was added and the reaction mixture was heated to 100°C for 30 minutes. The dolomite was filtered off with a cloth filter having fine pores and the organic material was washed with toluene (200 milliliters). Silica (50 grams) was added and the reaction mixture
5 was stirred for 30 minutes. The silica was filtered off with a cloth filter having fine pores and the organic material was washed with toluene (200 milliliters). Aqueous NH₃ (20% w/w)(250 milliliters) and N-acetylcysteine (12.5 grams) were added. The phases were separated. The organic phase was washed with HCl(10% w/w)(500 milliliters). The organic material was washed twice with
10 500 milliliters per wash of water. The toluene was distilled off under reduced pressure of about 120mbar. The pot temperature did not exceed 70°C. A dark brown viscous oil (about 500-700 milliliters) remained. To the hot mass in the pot, iso-butyl acetate (1162 milliliters) was added. A dark brown solution (about 1780 milliliters) formed.

15

A second reaction pot was loaded with heptane (7120 milliliters). Over a period of one hour, the contents of the first reaction pot were added to the second reaction pot. The precipitate was stirred for at least three hours and filtered off. The product was washed four times with 250 milliliters per wash of
20 heptane. The product was dried under reduced pressure of 40mbar at 80°C. The P2 Step (b) Product yield was 700 grams wet or 419 grams dry.

Analytical techniques including GPC, HPLC, and NMR were used to identify the product. GPC analysis showed: 1,3,5,7-tetrakis[3',4'-
25 (phenylethynyl)phenyl]adamantane (shown in Figure 1D) had a peak molecular weight of about 900; 1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-yl} benzene (shown in Figure 1F) had a peak molecular weight of about 1500; 1,3-bis{3'/4'-[1'',3'',5''-tris[3'''/4'''-(phenylethynyl)phenyl]adamant-7''-yl]phenyl}-5,7-bis[3''''/4''''-
30 (phenylethynyl)phenyl]adamantane (shown in Figure 1F) had a peak molecular weight of about 2100 (shoulder).

The melting point was 164-167°C. From NMR, a multiplet occurred at 6,9-8 ppm 2,8+-0,2H (aromatic part) and 1,7-2,7 ppm 1H+-0,2H (cage portion). From GPC, the ratio of the monomeric and small molecules to oligomeric compounds was 50±5%. FTIR showed the following:

5

PEAKS IN CENTIMETERS ⁻¹ (PEAK INTENSITY)	STRUCTURE
3050 (weak)	Aromatic C-H
2930 (weak)	Aliphatic C-H on adamantane
2200 (very weak)	Acetylene
1600 (very strong)	Aromatic C = C
1500 (strong)	
1450 (medium)	
1350 (medium)	

LC-MS study showed the presence of peaks:

- of main monomeric product m/z 840,
- of its derivatives, m/z 640, 740, 940; 706, 606, 806; 762, 662, 862; 938, 838, 1038
- of dimeric products, m/z 1402, 1302, 1502; 1326, 1226, 1426.

10

GPC 3 results follow.

Amount (Weight %)					Peak Molecular Weight (Relative to PS)				ΔR/R
Inc Mnmr	Mnmr	Dmr	Trmr	Hi Olgr	Inc Mnmr	Mnmr	Dmr	Trmr	Mnmr
7.7	36.4	26.8	12.3	16.8	546	763	1328	1672	0.14

15

Preparation 3

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Impact of Solvent on ratio of 1,3,5,7-tetrakis[3',4'-(phenylethynyl)phenyl]adamantane (shown in Figure 1D) to 1,3/4-bis{1',3',5'-tris[3''/4''-phenylethynyl)phenyl]adamant-7'-yl} benzene (shown in Figure 1F) and at least 1,3-bis{3'/4'-[1'',3'',5''-tris[3'''/4'''-(phenylethynyl)phenyl]adamant-7''-yl]phenyl}-5,7-bis[3''''/4''''-(phenylethynyl)phenyl]adamantane (shown in Figure 1F)

25

850 milliliters of P1 Step (a) Product was divided into four equal parts, and subjected to precipitation in petroleum ether, ligroine, heptane, and methanol. Each part was precipitated into 2520 ml of the solvent, vacuum
 5 filtered (Büchner funnel diam. 185 mm), washed on filter twice by 150 ml of the solvent, then dried in a vacuum oven for two hours at about 20°C, overnight at 40°C, and at 70-80°C to constant weight.

Precipitation into hydrocarbons resulted in very dispersed light beige
 10 powders that dried without complications. Precipitation into methanol gave heavy, brownish granular solid (particles size approximately 1 mm), which formed tar when dried at 20°C. This product was dried further.

Reaction mixtures were analyzed by GPC during the reaction and before
 15 precipitation. All filtrates and final solids were analyzed by GPC and the results are in Table 4. In Table 4, PPT stands for precipitation, monomer is 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane (shown in Figure 1A); dimer is 1,3/4-bis[1',3',5'-tris(3''/4''-bromophenyl)adamant-7'-yl]benzene (shown in Figure 1C); and trimer is 1,3-bis{3'/4'-[1'',3'',5''-tris(3'''/4'''-bromophenyl)adamant-
 20 7''-yl]phenyl}-5,7-bis(3''''/4''''-bromophenyl)adamantane (shown in Figure 1C).

Table 4

Peak Ratio [monomer to (dimer + trimer)] before PPT	Solvent For PPT	Peak Ratio [monomer to (dimer + trimer)] after PPT
75.0:25.0	Petroleum Ether	52.5:47.4
75.0:25.0	Ligroine	64.0:36.0
75.0:25.0	Heptane	66.2:33.8
75.0:25.0	Methanol	75.0:25.0

25 To summarize these results, the peak ratio of monomer to (dimer + trimer) in the reaction mixture was about 3:1. The product lost in hydrocarbons precipitation filtrates was mostly (>90%) monomer while losses in washing filtrates were negligible. There is no product in methanol precipitation filtrates.

The monomer to (dimer+trimer) ratio after precipitation increases (1:1→3:1), and monomer losses in the filtrates decrease (56→0%) in the sequence: petroleum ether, ligroine, heptane, and methanol.

5

Preparation 4 – P reparation of Thermosetting Component

The 1,3/4-bis{1',3',5'-tris[3''/4''-(phenylethynyl)phenyl]adamant-7'-yl} benzene (shown in Figure 1F) in the Preparation 1 product mixture is separated using preparative liquid chromatography (PLC). PLC is similar to the HPLC method described above but uses larger columns to separate larger quantities of the mixture (from several grams to several hundred grams).

10

Preparation 5 – P reparation of Thermosetting Component

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The 1,3-bis{3'/4'-[1'',3'',5''-tris[3'''/4'''-(phenylethynyl)phenyl]adamant-7''-yl]phenyl}-5,7-bis[3''''/4''''-(phenylethynyl) phenyl]adamantane (shown in Figure 1F) in the Preparation 1 product mixture is separated using preparative liquid chromatography (PLC).

20

Preparation 6 – P reparation of Thermosetting Component

The diamantane monomer of Formula V and oligomer or polymer of diamantane monomer of Formulae VI, VIII, X, XII, XV, XVI, and XVIII are prepared using the following method. As shown in Figure 2, diamantane is converted using bromine and a Lewis Acid catalyst to brominated diamantane product. The brominated diamantane product is then reacted with bromobenzene in the presence of a Lewis Acid catalyst to form bromophenylated diamantane. The bromophenylated diamantane is then reacted with a terminal alkyne in the presence of a catalyst system as used in the so-called Sonogashira coupling reaction. The product at each step is worked up as

25

30

described in our pending patent application PCT/US01/22204 filed October 17, 2001.

5 Preparation 7 – P reparation of Thermosetting Component (a)

 The diamantane monomer of Formula V and oligomer or polymer of
diamantane monomer of Formulae VI, VIII, X, XII, XV, XVI, and XVIII are
prepared using the following method. As shown in Figures 1A through 1F,
10 diamantane is converted to the bromophenylated compositions of diamantane
using similar synthetic procedures as described in Preparations 1 and 2. In
Figures 1A through 1C, diamantane is reacted with a substituted halogen phenyl
compound in the presence of a Lewis Acid catalyst as described in Preparations
1 and 2, and/or a second catalyst component as described in Preparation 2. A
15 mixture of monomers, dimers, trimers, and higher oligomers is obtained after
work-up of the reaction mixtures. In Figures 1D through 1F, the
bromophenylated diamantane mixture is then reacted with a terminal alkyne in
the presence of catalyst to produce the alkyne-substituted diamantane
compositions of the present invention.

20

Inventive Example 1

 In this example, the ethynyl containing group is first reacted with the
thermosetting component.

 To a 500-mL, 3 neck flask equipped with a condenser, a mechanical
25 stirrer and a nitrogen inlet-outlet were added thermosetting component similar to
Preparation 1 or 2 above (amount = 20.00 grams (20.19 millimoles));
dichlorobis(triphenylphosphine)palladium(II)(amount = 1.134 grams (1.62
millimoles)); triphenylphosphine (amount=0.848 gram (3.23 millimoles));
copper(I) iodide (amount=0.308 gram (1.62 millimoles)); triethylamine
30 (amount=70 milliliters); and toluene (amount=80 milliliters). The mixture was
heated to 80 °C and 4-ethynylaniline (amount=0.63 gram (5.2 millimoles)) in

20 milliliters of triethylamine were added to the reaction mixture dropwise. The reaction mixture was heated at 80 °C for 8 hours and then phenylacetylene (amount=16.50 grams (161.6 millimoles)) and triethylamine (amount=20 milliliters) were added to the reaction mixture dropwise. The solution was
5 heated at 80 °C for 8 hours.

The reaction mixture was cooled to room temperature and transferred to a 1 liter, 3 neck flask equipped with a condenser, a mechanical stirrer and a nitrogen inlet-outlet and toluene (100 milliliters) was added. The solution was then neutralized with 6N HCl. The resulting water was removed. The toluene
10 solution was then stirred with 100 mL of 6N HCl at 60 °C for 30 minutes. The mixture was filtered through celite® naturally occurring inorganic material. The aqueous solution was then removed. The HCl extraction was repeated for two more times. The toluene solution was then washed with 100 mL of deionized water twice. The solution was stirred with 100 mL of 0.1 M of N-acetyl-
15 cysteine in ammonia solution at 60 °C for 30 min. The aqueous solution was then removed. The ammonia extraction was repeated for five more times. The toluene was then removed by rotary evaporator and the resulting solid was dried under vacuum overnight to yield 17.10 grams (85.05%) of reddish solid (called Solid A).

20 To a 100-mL, 3 neck flask equipped with a condenser, a magnetic stirrer and a nitrogen inlet-outlet were added 20.70 mg (60% dispersion in mineral oil, which corresponds to 0.5176 mmol) of sodium hydride, and 20 ml of hexane. The mixture was stirred at room temperature for 5 minutes and upper hexane layer was decanted. To the above mixture were added tetrahydrofuran
25 (amount=20 milliliters)(THF) and 1.00 g of the above Solid A. The mixture was stirred at room temperature for 30 minutes and then epoxy functionalized polynorbornene (amount=0.6540 gram) was added. The solution was then heated at 65 °C for 12 hours. THF was then removed by rotary evaporator and the resulting mixture was dissolve in 15 ml of xylene (called Solution B). This
30 solution was washed by de-ionized water for 3 times. The preceding reaction scheme is shown in Figure 12 where although only 1,3/4-bis[1',3',5'-tris(3"/4"-

bromophenyl)adamant-7'-yl]benzene is shown, it is understood that similar reactions occur for 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane and 1,3-bis{3'/4'-[1'',3'',5''-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis{3'''/4'''-bromophenyl)adamantane.

5

To a 65-mL plastic bottle was added the above Solution B and ortho-cresol novolac (amount=0.030 gram; molecular weight of 1760; supplied by Schenectady International Inc.). The solution was stirred at room temperature for 1 hour. The solution was then filtered through a 0.1 μ m teflon filter.

10

The composition was applied to a substrate using typical coating conditions known to those skilled in the art. The resulting spun-on composition was baked for one minute under N₂ (< 50 ppm O₂) at each of the following temperatures: 125°C, 250°C, and 300°C. The furnace cure condition was 400°C for 60 minutes in N₂ (26 liters/minute) with ramping up from 250°C at 15 5°K per minute. The cure temperature range was from 350°C to 450°C. In each composition, the porogen decomposed and the decomposed porogen volatilized whereby pores formed in the composition. The layer had a refractive index of 1.433 and a thickness of 2414 Angstroms. Figure 13 shows the Scanning Electronic Microscope results.

20

Inventive Example 2:

In this example, the ethynyl containing group is first reacted with the porogen.

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To a 500-milliliter, 3 neck flask equipped with a condenser, a magnetic stirrer and a nitrogen inlet-outlet were added sodium hydride (amount=0.262 gram (60% dispersion in mineral oil, which corresponds to 6.54 millimoles)), and hexane (amount = 60 milliliters). The mixture was stirred at room temperature for 5 minutes and the upper hexane layer was decanted. To the above mixture were 30 added 4-ethynylaniline (amount=0.695 gram (5.93 millimoles)) and

tetrahydrofuran (THF, amount=144 grams). The solution was stirred at room temperature for 1 hour and epoxy functionalized polynorbornene (amount=15 grams) was added. The reaction mixture was heated at 60 °C for 12 hours. THF was then removed by rotary evaporator and the resulting mixture was dissolve
5 in 50 ml of toluene to form a solution (referred to below as Solution A).

To a 500-milliliter, 3 neck flask equipped with a condenser, a mechanical stirrer and a nitrogen inlet-outlet were added thermosetting component similar to Preparation 1 or 2 above (amount=25.75 grams (26.00 millimoles)),
10 dichlorobis(triphenylphosphine)palladium(II) (amount=1.461 grams(2.081 millimoles)), triphenylphosphine (amount=1.092 gram(4.162 millimoles)), copper(I) iodide (amount=0.3963 gram(2.081millimoles)), triethylamine (amount=160 milliliters), and toluene (amount=80 milliliters). The mixture was heated to 80 °C and the above Solution A was added to the reaction mixture
15 dropwise. The reaction mixture was heated at 80 °C for 12 hours and then phenylacetylene (amount=21.3 grams (208.1 millimoles)) and toluene (amount=30milliliters) were added to the reaction mixture dropwise. The solution was heated at 80 °C for 4 hours.

20 The reaction mixture was cooled to room temperature and transferred to a 1 liter, 3 neck flask equipped with a condenser, a mechanical stirrer and a nitrogen inlet-outlet and toluene (amount=100 milliliters) was added. The solution was then neutralized with 6N HCl. The resulting water was removed. The toluene solution was then stirred with 100 mL of 6N HCl at 60 °C for 30
25 min. The mixture was filtered through celite® naturally occurring inorganic material. The aqueous solution was then removed. The HCl extraction was repeated for two more times. The toluene solution was then washed with 100 mL of deionized water twice. The solution was stirred with 100 mL of 0.1 M of N-acetyl-cysteine in ammonia solution at 60 °C for 30 min. The aqueous
30 solution was then removed. The ammonia extraction was repeated for five more times. The toluene was then removed by rotary evaporator and the resulting

solid was dried under vacuum overnight. The preceding reaction scheme is shown in Figure 14 where although only 1,3/4-bis[1',3',5'-tris(3"/4"-bromophenyl)adamant-7'-yl]benzene is shown, it is understood that similar reactions occur for 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane and 1,3-bis{3'/4'-[1",3",5"-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis{3'''/4'''-bromophenyl)adamantane.

To a 125-milliliter plastic bottle were added 4.48 g of the above solid, 0.047 g of ortho-cresol novolac (amount=0.047 gram; molecular weight of 1760; supplied by Schenectady International Inc.) and xylenes (40.74 grams). The solution was stirred at room temperature for 1 hour. The solution was then filtered through a 0.1 μm teflon filter.

The composition was applied to a substrate using typical coating conditions known to those skilled in the art. The resulting spun-on composition was baked for one minute under N_2 (< 50 ppm O_2) at each of the following temperatures: 125°C, 250°C, and 300°C. The furnace cure condition was 400°C for 60 minutes in N_2 (26 liters/minute) with ramping up from 250°C at 5°K per minute. The cure temperature range was from 350°C to 450°C. In each composition, the porogen decomposed and the decomposed porogen volatilized whereby pores formed in the composition. After bake, the layer had a refractive index of 1.636 and a thickness of 1255 Angstroms. After cure, the layer had a refractive index of 1.398 and a thickness of 1056 Angstroms. Figure 15 shows Scanning Electron Microscope results.

Inventive Example 3:

In this example, the ethynyl containing group is first reacted with the porogen.

To a 300-milliliter, 3 neck flask equipped with a condenser, a magnetic stirrer and a nitrogen inlet-outlet were added triphenylphosphine (amount=9.76 grams (37.2 millimoles)), diethyl azodicarboxylate (amount=6.49 grams (37.2

millimoles)), 3-hydroxyphenylacetylene (amount=4.00 grams (33.9 millimoles)), and tetrahydrofuran (THF; amount=90 milliliters). A clear solution was obtained after the mixture was stirred at room temperature for 5 minutes. To this solution was then added 10.76 g of polycaprolactone in 40 ml of THF solution dropwise at room temperature. The solution was stirred at room temperature for 12 hours. THF was then partially removed by rotary evaporator to make a 40 ml viscous solution and ethyl ether (amount=50 milliliters) was added to the mixture and put into refrigerator for 30 min. The precipitate that formed was removed by filtration. Ethyl ether in the filtrate was then removed by rotary evaporator. To this viscous solution was added methylene chloride (50 milliliters) and put into refrigerator overnight. The precipitation that formed was removed by filtration. The solvent was then removed by rotary evaporator to yield a viscous liquid (referred to below as Liquid A).

To a 500-mL, 3 neck flask equipped with a condenser, a mechanical stirrer and a nitrogen inlet-outlet were added thermosetting component similar to Preparation 1 or 2 above (amount=13.34 grams (13.47 millimoles)), dichlorobis(triphenylphosphine)palladium(II) (amount=0.7566 gram (1.078 millimoles)), triphenylphosphine (amount=0.5655 gram (2.156 millimoles)), copper(I) iodide (amount=0.2053 gram (1.078 millimoles)), triethylamine (amount=110 milliliters). The mixture was heated to 80 °C and 12.30 g of the above Liquid A with toluene (amount=40 milliliters) was added to the reaction mixture dropwise. The reaction mixture was heated at 80 °C for 12 hours and then 10.32 g (101.1 mmol) of phenylacetylene and toluene (amount=30 milliliters) were added to the reaction mixture dropwise. The solution was heated at 80 °C for 4 hours.

The reaction mixture was cooled to room temperature and transferred to a 1 liter, 3 neck flask equipped with a condenser, a mechanical stirrer and a nitrogen inlet-outlet and toluene (amount=100 milliliters) was added. The solution was then neutralized with 6N HCl. The resulting water was removed. The toluene solution was then stirred with 100 mL of 6N HCl at 60 °C for 30

min. The mixture was filtered through celite®. The aqueous solution was then removed. The HCl extraction was repeated for two more times. The toluene solution was then washed with 100 mL of deionized water twice. The solution was stirred with 100 mL of 0.1 M of N-acetyl-cysteine in ammonia solution at 5 60 °C for 30 min. The aqueous solution was then removed. The ammonia extraction was repeated for five more times. The toluene was then removed by rotary evaporator and the resulting solid was dried under vacuum overnight. The preceding reaction scheme is shown in Figure 16 where although only 1,3/4-bis[1',3',5'-tris(3"/4"-bromophenyl)adamant-7'-yl]benzene is shown, it is 10 understood that similar reactions occur for 1,3,5,7-tetrakis(3'/4'-bromophenyl)adamantane and 1,3-bis{3'/4'-[1",3",5"-tris(3'''/4'''-bromophenyl)adamant-7''-yl]phenyl}-5,7-bis{3''''/4''''-bromophenyl)adamantane.

15 To a 125-milliliter plastic bottle were added 2 grams of the above solid, polycarbosilane $(CH_2SiH_2)_q$ where q is 20-30 (amount=0.1334gram; supplied by Starfire Systems, Inc.) and 20 g of xylenes (amount=20 grams). The solution was heated at 145 °C for 15.5 hours. The solution was then filtered through a 20 0.1 µm teflon filter.

The composition was applied to a substrate using typical coating conditions known to those skilled in the art. The resulting spun-on composition was baked for one minute under N_2 (< 50 ppm O_2) at each of the following 25 temperatures: 125°C, 250°C, and 300°C. The furnace cure condition was 400°C for 60 minutes in N_2 (26 liters/minute) with ramping up from 250°C at 5°K per minute. The cure temperature range was from 350°C to 450°C. In each composition, the porogen decomposed and the decomposed porogen volatilized whereby pores formed in the composition. After bake, the layer had 30 a refractive index of 1.617 and a thickness of 5640 Angstroms. After cure, the layer had a refractive index of 1.593 and a thickness of 3784 Angstroms. The above formulation without the adhesion promoter had an after base refractive

index of 1.639, an after bake thickness of 1369 Angstroms, an after cure refractive index of 1.584, an after cure thickness of 993 Angstroms, a degassed dielectric constant of 2.66, and an additional cure refractive index of 1.562.